

Supplementary Materials

Characterization and Syngas Production at Low Temperature via Dry Reforming of Methane over Ni-M (M = Fe, Cr) Catalysts Tailored from LDH Structure

Manel Hallassi ^{1,2,3}, Rafik Benrabaa ^{2,3}, Nawal Fodil Cherif ⁴, Djahida Lerari ⁴, Redouane Chebout ⁴, Khaldoun Bachari ⁴, Annick Rubbens ⁵, Pascal Roussel ⁵, Rose-Noëlle Vannier ⁵, Martine Trentesaux ⁵ and Axel Lôfberg ^{5,*}

¹ Université 20 Août-Skikda, Faculté de Technologie, Département de Génie des Procédés, BP 26, route Al-Hadaiek, 21000 Skikda, Algérie

² Laboratoire de Physico-Chimie des Matériaux, Faculté des Sciences et de la Technologie, Université Chadli Bendjedid-El Tarf B.P 73, El Tarf 36000, Algérie

³ Laboratoire de Matériaux Catalytiques et Catalyse en Chimie Organique, Faculté de Chimie, USTHB, BP 32, El-Alia, 16111 Bab Ezzouar, Alger, Algérie

⁴ Centre de Recherche Scientifique et Technique en Analyses Physico-Chimiques, BP 384, Siège ex-Pasna Zone Industrielle, Bou-Ismaïl CP 42004, Tipaza, Algérie

⁵ Univ. Lille, CNRS, Centrale Lille, ENSCL, Univ. Artois, UMR 8181 - UCCS - Unité de Catalyse et Chimie du Solide, F-59000 Lille, France

* Correspondence: axel.lofberg@univ-lille.fr; Tel.: +33-03-20-43-45-27

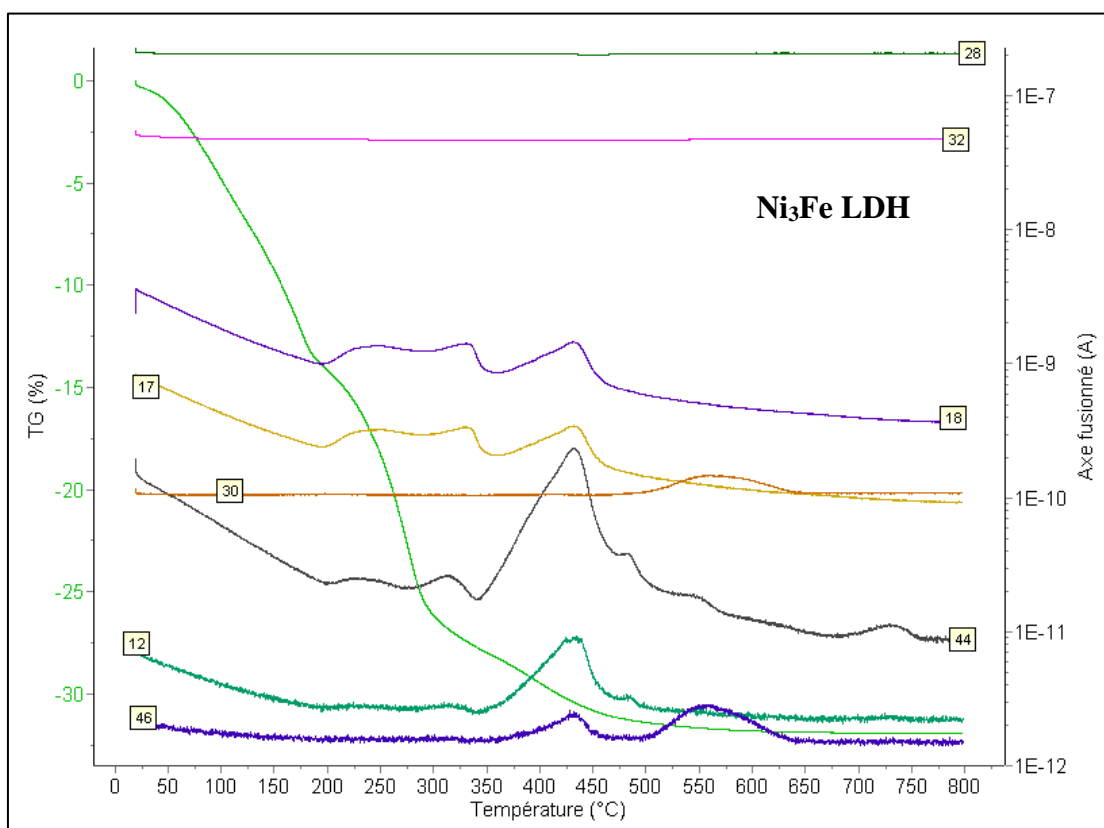
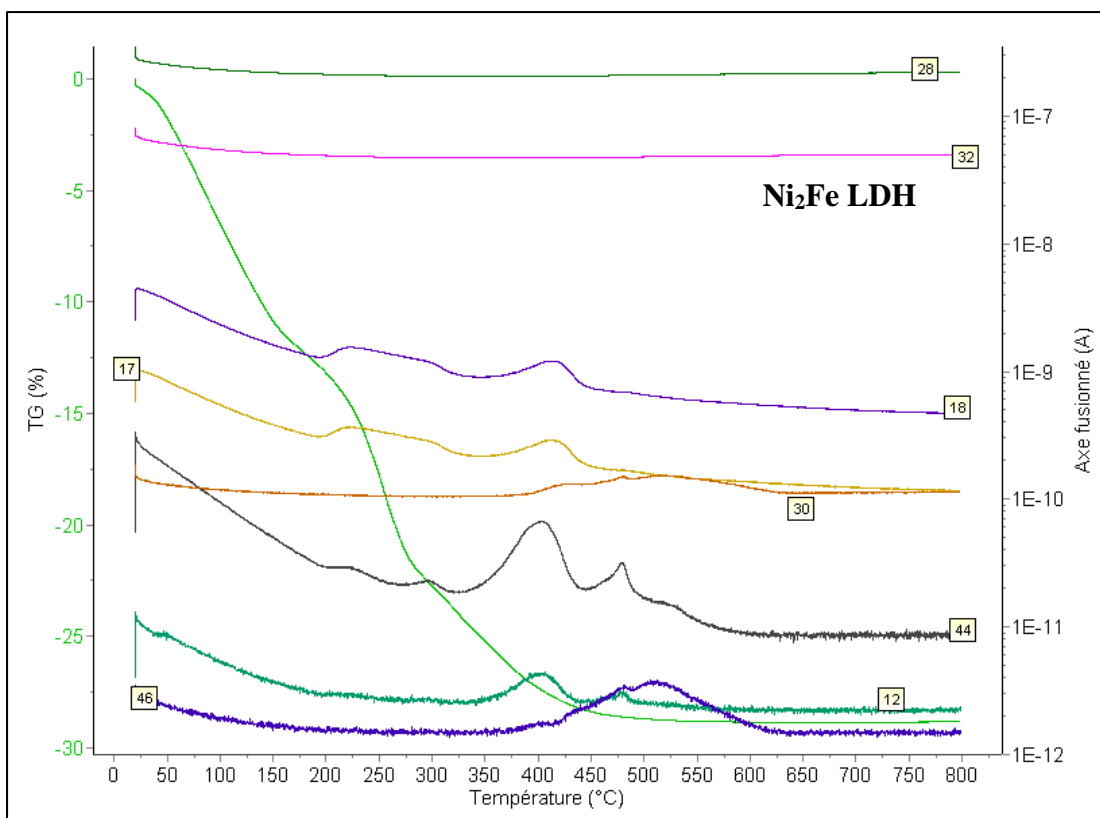


Figure S1. TG-MS curves of Ni₂Fe and Ni₃Fe LDH precursors performed in air atmosphere.

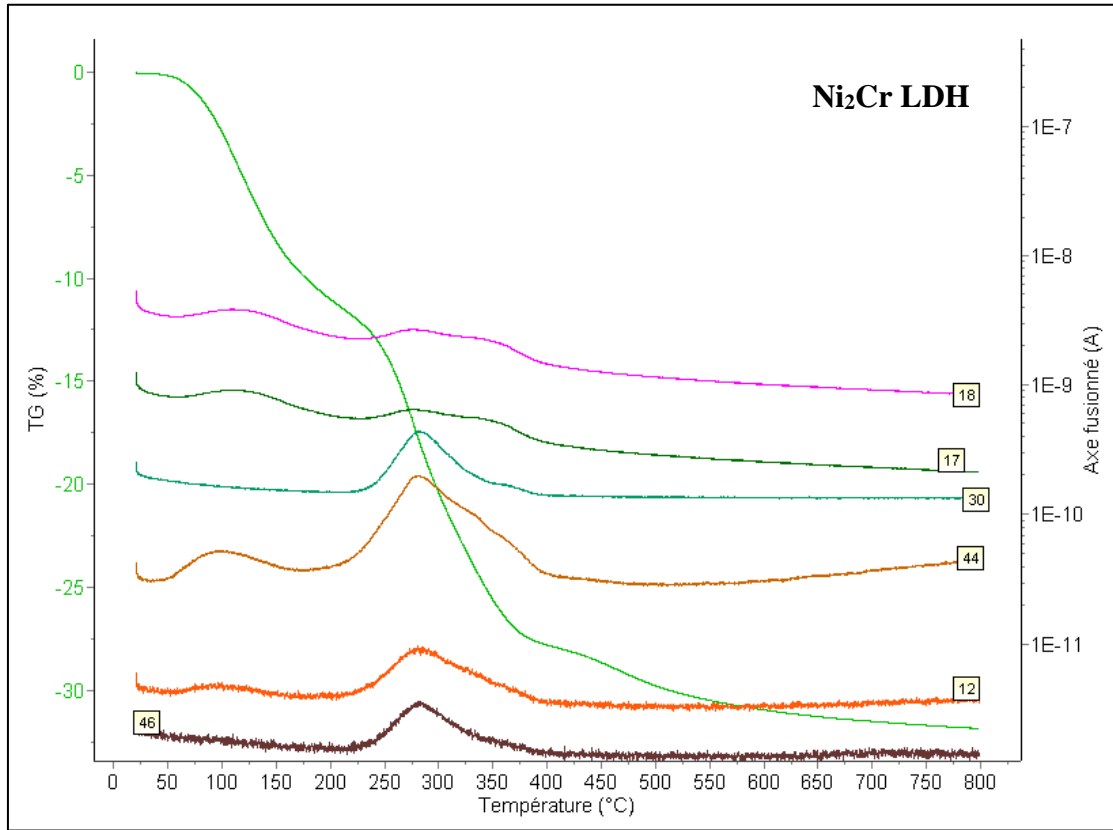


Figure S2. TG-MS curves of $\text{Ni}_2\text{Cr LDH}$ precursor performed in air atmosphere.

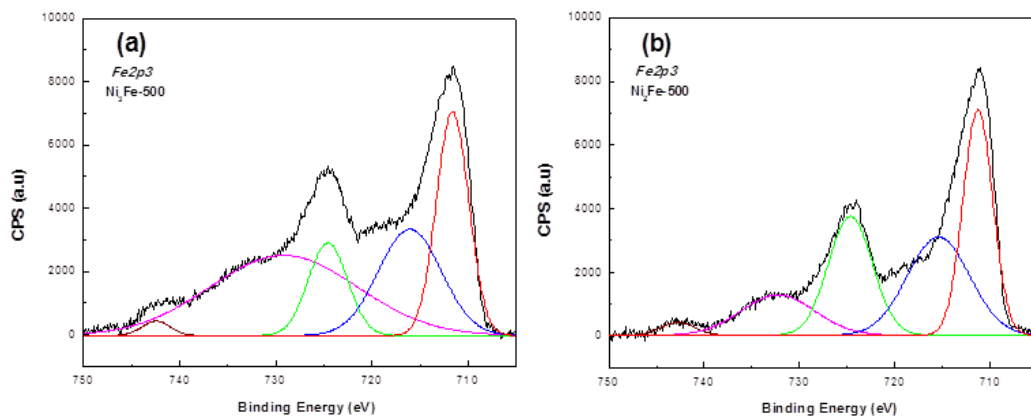


Figure S3. XPS spectra of $Fe2p_{3/2}$ species of (a) $Ni_3Fe-500$ and (b) $Ni_2Fe-500$.

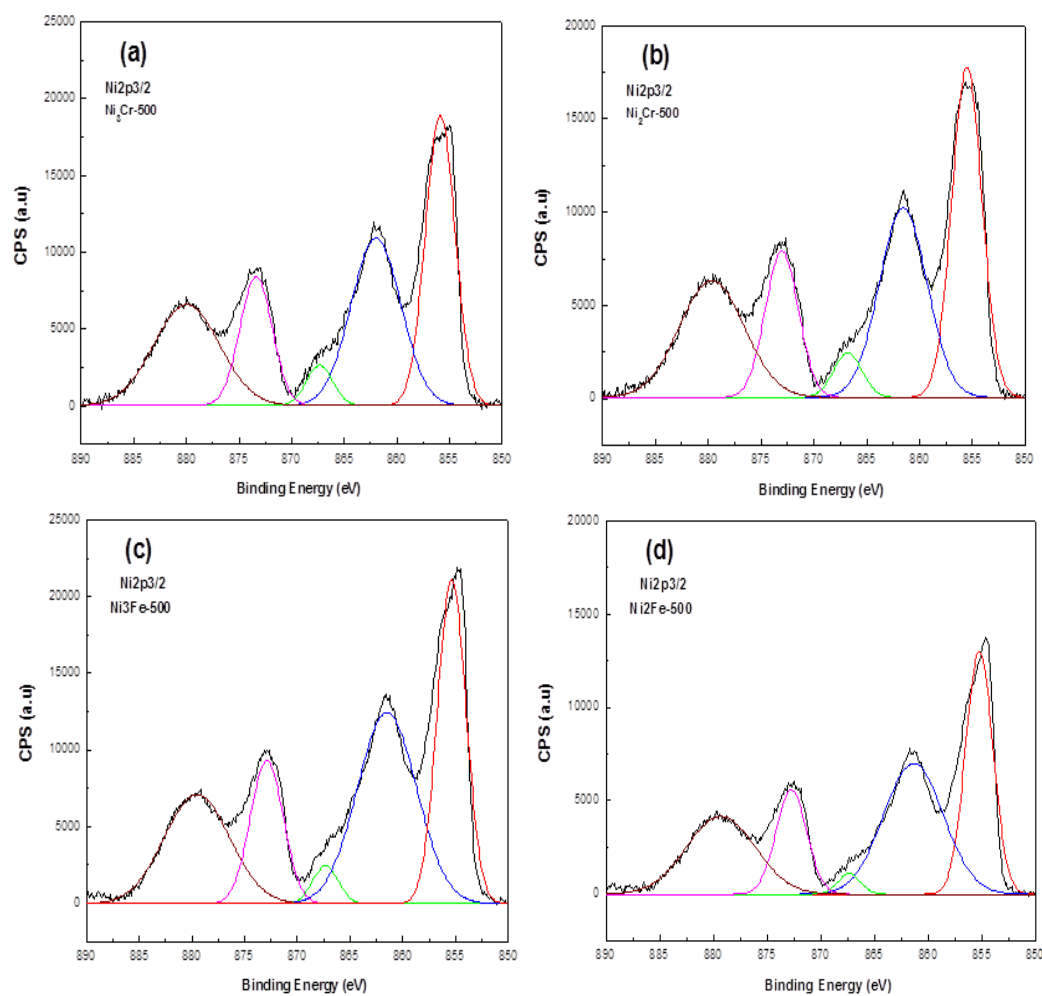


Figure S4. XPS spectra of $Ni2p_{3/2}$ species of (a) $Ni_3Cr-500$, (b) $Ni_2Cr-500$, (c) $Ni_3Fe-500$ and (d) $Ni_2Fe-500$.

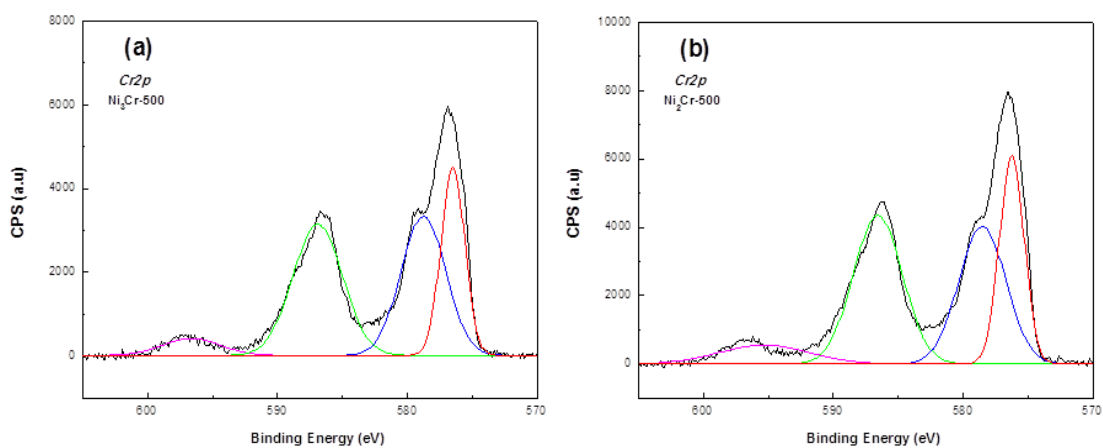


Figure S5. XPS spectra of Cr2p species of (a) Ni₃Cr-500 and (b) Ni₂Cr-500.

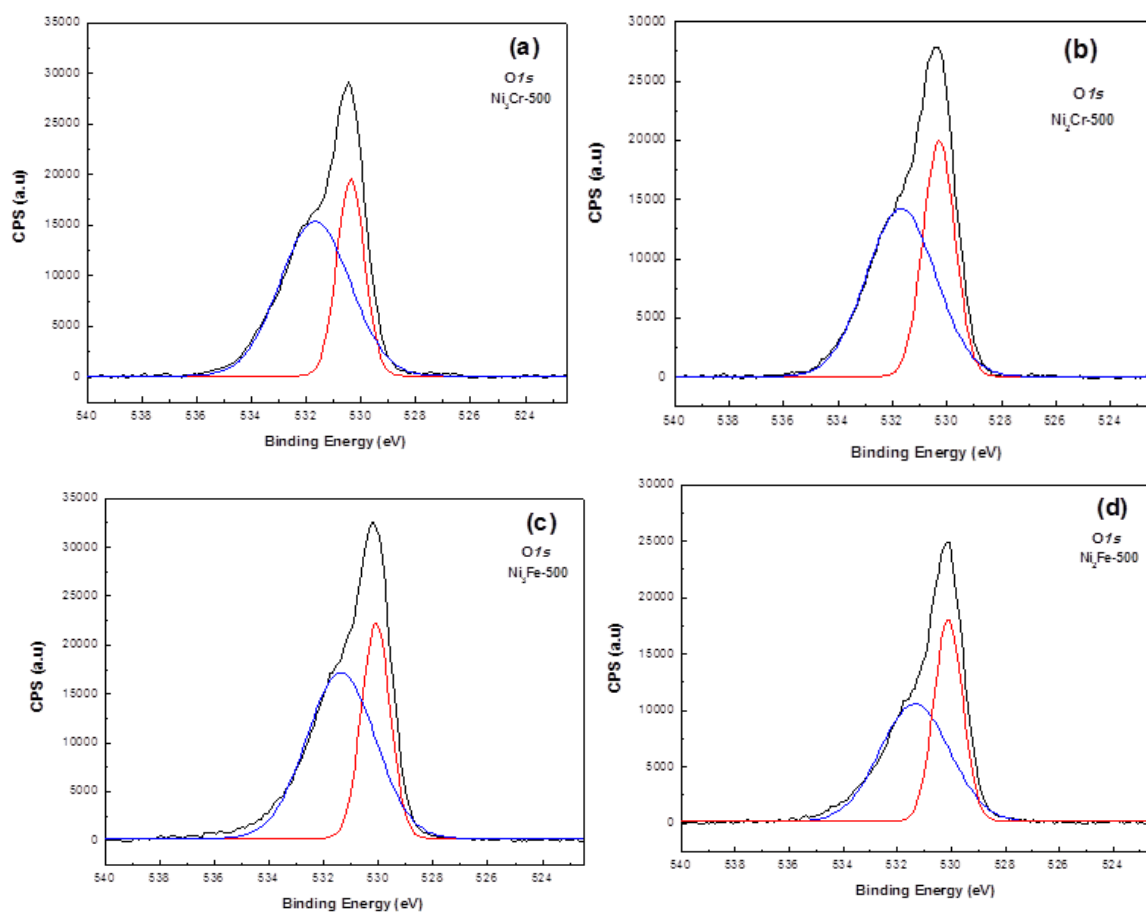


Figure S6. XPS spectra of O1s species of (a) Ni₃Cr-500, (b) Ni₂Cr-500, (c) Ni₃Fe-500 and (d) Ni₂Fe-500.

Table S1. TGA-MS of Ni_RFe and Ni_RCr LDH precursors.

LDH	Domain of T° (°C)	Species released	% of lost mass	% total lost mass	% total theoretical lost mass
R=2 [Ni _{0.66} Fe _{0.33} (OH) ₂] (CO ₃) _{0.165}	30 -150	H ₂ O	6.02	24,01	24,99
	150-281	H ₂ O,NO,	10.97		
	281 -474	CO ₂ , NO ₂ NO,CO ₂ NO ₂	7.02		
R=3 [Ni _{0.75} Fe _{0.25} (OH) ₂] (CO ₃) _{0.125}	30 -16	H ₂ O	9,25	27,44	23,34
	164-290	H ₂ O,NO,	12,75		
	290 -480	CO ₂ NO ₂ NO,CO ₂ NO ₂	5,44		
R=2 [Ni _{0.66} Cr _{0.33} (OH) ₂] (CO ₃) _{0.165}	53-156	H ₂ O	9,37	31,73	25,3
	156-345	H ₂ O,NO,	18,62		
	345 -570	CO ₂ NO ₂ NO,CO ₂ NO ₂	3,74		
R=3 [Ni _{0.75} Cr _{0.25} (OH) ₂] (CO ₃) _{0.125}	51 -159	H ₂ O	9,17	31,53	23,85
	159-352	H ₂ O,NO,	18,62		
	352 -580	CO ₂ NO ₂ NO,CO ₂ NO ₂	3,74		

Table S2. Catalytic performances in DRM, temperature programmed mode

Catalysts	T (°C)	X% CH₄	X% CO₂	S% H₂	H₂/CO
Ni ₂ Cr-500	400	0	0	0	0
	450	0	0	0	0
	500	16	18	59	0.6
	550	37	36	75	0.9
	600	63	48	73	1.2
	650	46	28	55	1.3
Ni ₃ Cr-500	400	1	1	0	0
	450	4	4	36	0.3
	500	23	22	68	0.7
	550	40	29	70	1.1
	600	47	30	52	1.8
	650	70	27	25	0.9
Ni ₂ Fe-500	400	3	1	2	0
	450	3	1	2	0
	500	4	1	2	0
	550	7	1	2	0
	600	8	1	5	0.4
	650	6	1	8	0.5
Ni ₃ Fe-500	400	2	1	1	0
	450	2	1	1	0
	500	3	1	1	0.5
	550	3	1	7	0.6
	600	5	1	6	0.7
	650	7	2	4	0.3