



HAL
open science

Archaerhodopsin 3 is an ideal template for the engineering of highly fluorescent optogenetic reporters

Krystyna Herasymenko, Danushka Walisinghe, Masae Konno, Leonardo Barneschi, Isabelle de Waele, Michel Sliwa, Keiichi Inoue, Massimo Olivucci, Stefan Haacke

► To cite this version:

Krystyna Herasymenko, Danushka Walisinghe, Masae Konno, Leonardo Barneschi, Isabelle de Waele, et al.. Archaerhodopsin 3 is an ideal template for the engineering of highly fluorescent optogenetic reporters. *Chemical Science*, 2024, *Chem Sci*, 16 (2), pp.761-774. 10.1039/d4sc05120c . hal-04844689

HAL Id: hal-04844689

<https://hal.univ-lille.fr/hal-04844689v1>

Submitted on 22 Jan 2025

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



Distributed under a Creative Commons Attribution 4.0 International License

Cite this: *Chem. Sci.*, 2025, 16, 761

All publication charges for this article have been paid for by the Royal Society of Chemistry

Archaerhodopsin 3 is an ideal template for the engineering of highly fluorescent optogenetic reporters†

Krystyna Herasymenko,^{‡a} Danushka Walisinghe,^{‡b} Masae Konno,^{‡c} Leonardo Barneschi,^d Isabelle de Waele,^e Michel Sliwa,^{ef} Keiichi Inoue,^{‡c} Massimo Olivucci^{‡*bd} and Stefan Haacke^{‡*a}

Archaerhodopsin-3 (AR-3) variants stand out among other rhodopsins in that they display a weak, but voltage-sensitive, near-infrared fluorescence emission. This has led to their application in optogenetics both in cell cultures and small animals. However, in the context of improving the fluorescence characteristics of the next generation of AR-3 reporters, an understanding of their ultrafast light-response in light-adapted conditions, is mandatory. To this end, we present a combined experimental and computational investigation of the excited state dynamics and quantum yields of AR-3 and its DETC and Arch-5 variants. The latter always display a mixture of all-*trans*/15-*anti* and 13-*cis*/15-*syn* isomers, which leads to a bi-exponential excited state decay. The isomerisation quantum yield is reduced more than 20 times as compared to WT AR-3 and proves that the steady-state fluorescence is induced by a single absorption photon event. In wild-type AR-3, we show that a 300 fs, barrier-less and vibrationally coherent isomerization is driven by an unusual covalent electronic character of its all-*trans* retinal chromophore leading to a metastable twisted diradical (TIDIR), in clear contrast to the standard charge-transfer scenario established for other microbial rhodopsins. We discuss how the presence of TIDIR makes AR-3 an ideal candidate for the design of variants with a one-photon induced fluorescence possibly reaching the emission quantum yield of the top natural emitter neorhodopsin (NeoR).

Received 31st July 2024

Accepted 13th November 2024

DOI: 10.1039/d4sc05120c

rsc.li/chemical-science

1 Introduction

In spite of its tiny fluorescence quantum yield (Φ_f), archaerhodopsin-3 (AR-3) from the halobacterium *Halorubrum sodomense*, has been successfully used as fluorescent genetically encoded voltage indicator (GEVIs), thus enabling the visualization of neuronal activity with spatiotemporal precision.^{1–4} To enhance the relatively low Φ_f of AR-3 and expand its applicability as GEVI, several AR-3 variants were designed over the past

years by screening libraries with randomly mutated AR-3.^{5–8} However, the highest Φ_f reported for the seven-fold mutated ARCH-7 is only 1.2%,⁵ still much lower than the *ca.* 20% value recently found for the exceptional naturally fluorescent neorhodopsin (NeoR).^{9,10} There is therefore a need to understand if AR-3 can be engineered to generate variants approaching or overcoming, the NeoR limit or if, alternatively, there are fundamental reasons why this is not possible.

A first step towards an understanding of AR-3 was achieved by some of us *via* a theoretical/computational study of the electronic structure of the first excited state (S_1) of its retinal protonated Schiff base (RPSB) chromophore (and fluorophore).¹¹ A schematic representation of the established fluorescence mechanism in AR-3 mutants is given in Fig. 1. A potential energy surface (PES) barrier along the S_1 isomerization path of RPSB was found to control the lifetime of the fluorescent state (FS) and prevent its efficient decay into the region of an S_1/S_0 conical intersection (CoIn). By computing the height of the barrier ($E_{S_1}^f$) for different AR-3 mutants, a correlation was established between barrier height and mutation-dependent Φ_f .¹¹ This result indicated that: (i) the higher fluorescence of randomly engineered variants is due to a 1-photon excitation process that is ineffective in WT AR-3 (ref. 12 and 13) and (ii) the increase in 1-photon Φ_f correlates with the stability ($E_{TIDIR-FS}$) of

^aUniversity of Strasbourg, CNRS, IPCMS, 23 Rue du Loess, 67034 Strasbourg, France. E-mail: stefan.haacke@ipcms.unistra.fr

^bDepartment of Chemistry, Bowling Green State University, Bowling Green, OH 43403, USA. E-mail: molivuc@bgsu.edu

^cThe Institute for Solid State Physics, University of Tokyo 5-1-5, Kashiwanoha, Kashiwa, Chiba, 277-8581, Japan. E-mail: inoue@issp.u-tokyo.ac.jp

^dDipartimento di Biotecnologie, Chimica e Farmacia, Università di Siena, I-53100 Siena, Italy

^eLASIRE, Université de Lille, CNRS, 59000, Lille, France

^fLOB, CNRS, INSERM, École Polytechnique, Inst. Polytechnique de Paris, 91120 Palaiseau, France

† Electronic supplementary information (ESI) available: Experimental and computational methods, additional data and figures femtosecond spectroscopy, protocol light adaptation, etc. See DOI: <https://doi.org/10.1039/d4sc05120c>

‡ These authors contributed equally.



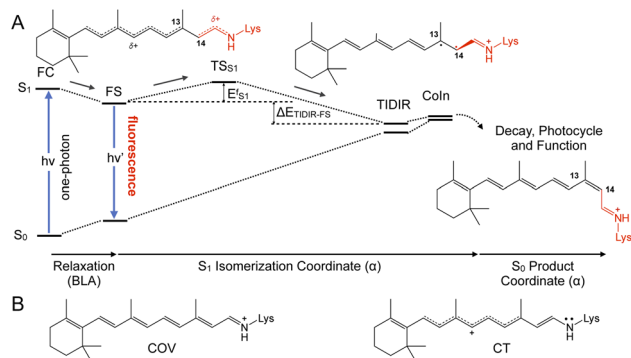


Fig. 1 (A) Representation of the chromophore isomerization path. FS corresponds to the fluorescent state. TIDIR represents the photoisomerization channel located near Coln. Lewis formulas representing product, FS and TIDIR display distinct degrees of double bond twisting and charge transfer. (B) Representation of electronic characters of the chromophore COV vs. CT. Note the different bond length alternation (BLA).

an unusual S_1 twisted bi-radical (TIDIR) of RPSB displaying a covalent (COV) electronic character, in stark contrast to the dominant charge transfer (CT) character of the S_1 state in other microbial rhodopsins.^{14–19} Remarkably, in a more recent computational study the same mechanism was shown to be responsible for the record Φ_f of NeoR.^{10,20}

Two central questions remain to be answered to establish WT AR-3 as a suitable starting point for fluorescent GEVI engineering. The first is whether the transient TIDIR intermediate is also a characteristic of WT AR-3 in spite of a substantial absence of 1-photon fluorescence in this species (WT AR-3 fluorescence originates from a complex 3-photon process).^{12,13} The second is whether it is possible to provide direct or indirect experimental evidence for the existence of TIDIR. Experimental studies on the fluorescence lifetimes of QuasAr1, QuasAr2 and Archon1 and its mutants were recently reported,^{21–23} confirming that the fluorescence is induced by a 1-photon process. Importantly, in contrast with the fact that light-adapted (LA) WT AR-3 contains a single all-*trans* chromophore isomer, the chromophore of the variants mentioned above was found to correspond to a mixture of all-*trans* and 13-*cis* RPSBs.^{22,23} For Archon1, it was also argued that the trans-membrane voltage acts on the ground isomer composition, and hence on the Φ_f .²²

In the present paper, we address the two questions above *via* a combination of spectroscopic and non-adiabatic molecular dynamics studies. Experimentally, we study the transient spectroscopy and photoisomerization efficiency of WT AR-3 and, additionally, of two specific AR-3 mutants: the double mutant DETC (D95E, T99C) and the five-fold mutant ARCH-5 (V59A, P60L, D95E, T99C, P196S) initially reported by McIsaac *et al.*⁵ For WT AR-3 at pH6, the excited state lifetimes (τ_{S_1}) and isomerization quantum yield (Φ_{iso}) were measured for the first time, providing indirect evidence for an excited state barrier ($E_{S_1}^\ddagger$) induced by the TIDIR. Computationally, we focus on the simulation of the photoinduced dynamics of the WT AR-3 using a set of 200 quantum-classical trajectories. The results consistently support the existence of a TIDIR intermediate indicating

that the mechanism illustrated in Fig. 1 is general and valid for WT AR-3, its variants as well as for NeoR. Accordingly, one concludes that the problem of finding highly fluorescent AR-3 variants, namely by increasing the average excited state lifetime (τ) of the system, can be solved by finding mutations that sufficiently destabilize TIDIR with respect to FS. This conclusion is indirectly supported by the spectroscopic data of the two AR-3 mutants, as we report that the $\langle \tau \rangle$ dramatically increases to 29 and 66 ps for DETC and ARCH-5, respectively, and that the Φ_{iso} is reduced by more than 20 times.

2 Experimental results and discussion

2.1 Retinal configuration analysis in dark- and light-adapted proteins determined by HPLC

High-performance liquid chromatography (HPLC) analysis was used to investigate the isomeric configurations of the RPSB in

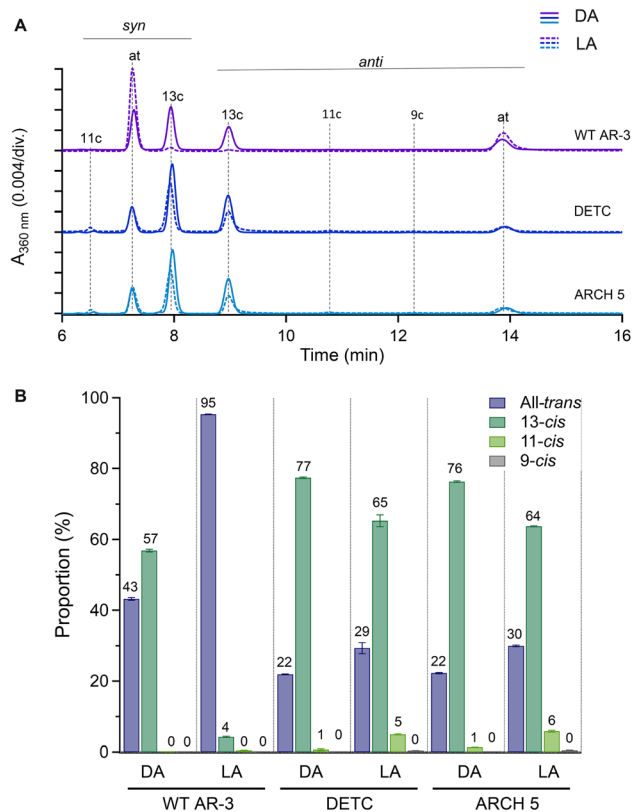


Fig. 2 HPLC analysis of the retinal chromophore of AR-3 WT, DETC and ARCH-5 in the dark-adapted (DA) and the light-adapted (LA) states. (A) Representative HPLC chromatogram of the retinal oxime in dark-adapted (DA, solid line) and light-adapted (LA, dotted line) states. The retinal oxime was produced by the hydrolysis reaction between the retinal chromophore and hydroxylamine. *Syn* and *anti* refer to the configuration of the C=N bond of the product retinal oxime, which does not represent the configuration of the retinal Schiff base in the original proteins. The retention time of each isomer was normalized based on that of each isomer in DA. (B) The ratio of each isomer (mean \pm standard deviation, $n = 3$ independent experiments). Sample illumination and HPLC analysis were performed three times independently. The numbers at the top of each bar indicate the composition of each isomer.



light- and dark-adapted forms of WT AR-3 and the two mutants (*vide supra*). The results are given in Fig. 2. For the light-adaptation procedure, details are given in Fig. S8† together with the common associated red-shifts of absorption spectra, indicating accumulation of all-*trans* RPSB. In all proteins, the retinal primarily existed in all-*trans*/15-*syn* (AT) and 13-*cis*/15-*anti* (13-C) forms. In the dark-adapted state, 43 and 57% of WT AR-3 had AT and 13-C retinal, respectively, whereas 95% of the chromophore was in the AT form in the light-adapted state like in br.²⁴ For DETC and ARCH-5, the dark-adapted state contained ~80% of the retinal in the 13-C form. Upon light adaptation, although a slight increase in the AT form was observed, but the 13-C remains in majority, accounting for ~63% and ~65% of the retinal in DETC and ARCH-5, respectively.

2.2 Raman spectroscopy of light-adapted proteins

To investigate the retinal isomer composition in the LA WT AR-3, DETC and ARCH-5 Fourier transform Raman spectroscopy was used. Light adaptation was performed with the same protocol as for the HPLC experiments above (Fig. S8†). Fig. 3 presents the obtained Raman spectra for each of the proteins. For the fingerprint region (C–C single bond stretching) we can see that the WT has predominantly all-*trans* configuration of the retinal chromophore (1169 cm⁻¹ and 1202 cm⁻¹ bands).²⁵ On the other hand, in DETC, a peak at 1186 cm⁻¹, previously assigned to the 13-*cis*, 15-*syn* configuration,^{25,26} is observed. In ARCH-5, both peaks are unresolved, which could indicate a slight red-shift of the 13-*cis*-related vibration in this mutant. Moreover, in the methyl group rocking region around 1000 cm⁻¹, WT and ARCH-5 display a single peak at 1008 cm⁻¹, while DETC shows an additional peak at 1000 cm⁻¹, which could also be a sign of the existence of two isomers. However, in the hydrogen-out-of-plane (HOOP) region around 800 cm⁻¹ that was previously reported to be an indicator of a distorted 13-*cis*

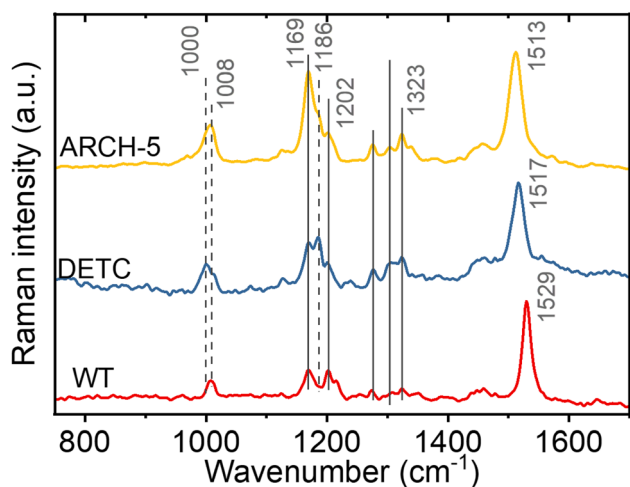


Fig. 3 FT-Raman spectroscopy. Comparison of the off-resonance FT-Raman spectra of light-adapted WT AR-3, DETC and ARCH-5 proteins at pH 6. The positions of the common bands are highlighted with solid vertical lines. The positions of the peaks that shifted or were not present in one of the proteins are indicated by the dashed lines.

isomer²⁶ no peak is observed for any of the three proteins. In addition, the red-shift of the C=C double frequency (from 1529 cm⁻¹ in WT to 1513 cm⁻¹ in ARCH-5) is observed for mutants, in line with the red shift of the mutants' absorption spectra with respect to the WT.

In conclusion, the isomer composition of the light-adapted proteins determined by Fourier transform Raman spectroscopy is qualitatively in agreement with the more precise and quantitative HPLC method (*vide supra*), with the additional information that the 13-*cis*, 15-*syn* isomer appears to be planar since no HOOP activity is observed.

2.3 Excited state lifetimes

To study the impact of point mutations on the *in vitro* fluorescence properties, the Φ_f and fluorescence lifetimes were studied for DETC and ARCH-5 in their LA forms. The absorption and emission spectra of these mutants were measured (Fig. 4). These are known to be red-shifted with respect to the WT.⁵ Moreover, Φ_f for DETC and ARCH-5 at pH 6 were calculated as described in Methods.† The obtained values of $3.6 \pm 0.6 \times 10^{-3}$ and $9 \pm 1 \times 10^{-3}$ for DETC and ARCH-5 respectively are in agreement with the ones previously obtained for slightly lower pH values.⁵

Broadband fluorescence up-conversion spectroscopy (FLUPS)^{27,28} with ~150 fs time resolution (see Methods†) was used to investigate the fluorescence lifetimes of LA WT AR-3 and the two mutants at pH 6. Fig. 5A presents an example of the time evolution of DETC fluorescence spectra. According to this figure, the spectral shape does not change with time, only the fluorescent intensity decays. This indicates that the excited state dynamics do not depend on the detection wavelength and that both single and integrated kinetic analyses are relevant.

For an analysis of the kinetics, a spectral range of 15 nm around the fluorescence maximum (WT 722 nm, DETC and ARCH-5 735 nm) was averaged. The panel B of Fig. 5 presents experimental data (dots) for normalized kinetic traces for WT AR-3 (blue) and its mutants DETC (red) and ARCH-5 (yellow). It is clearly seen, that the WT fluorescence decays in a few ps while the mutants' fluorescence lives significantly longer. The kinetics have bi-exponential decay character for all the samples (Fig. S2†). Solid lines display the results of the multi-exponential fit. A Gauss function (representing the instrument

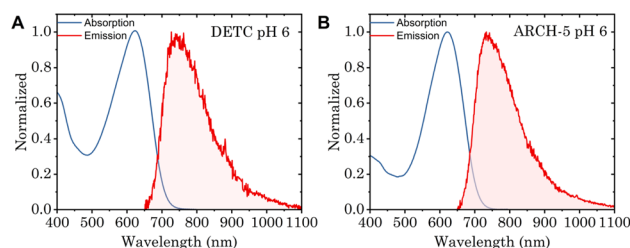


Fig. 4 (A) Normalized absorption (blue) and emission (red) spectra of DETC at pH 6, excitation wavelength 615 nm. (B) Normalized absorption (blue) and emission (red) spectra of ARCH-5 at pH 6, excitation wavelength 600 nm.



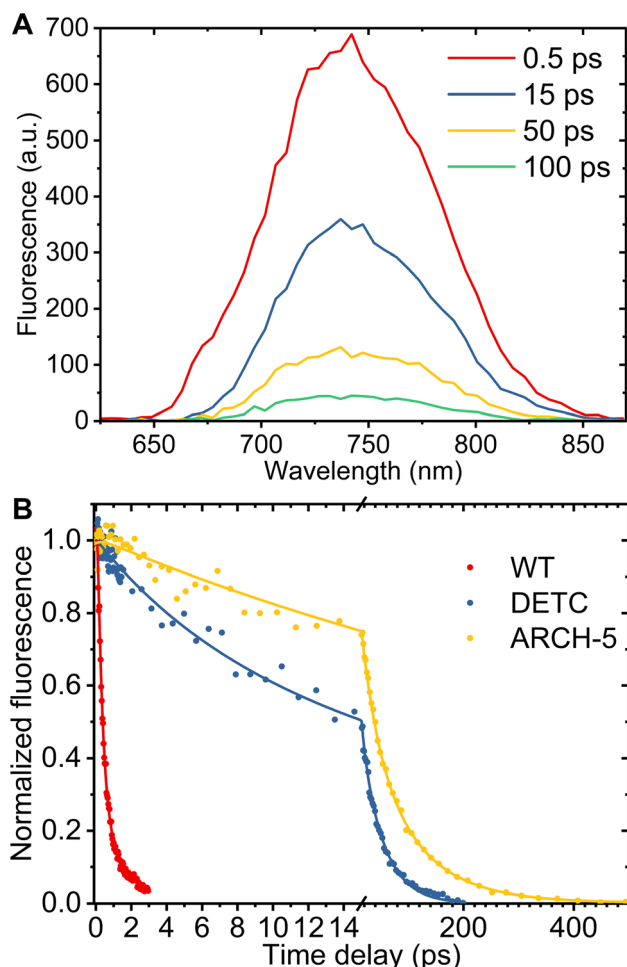


Fig. 5 Results of FLUPS experiments. (A) Fluorescence spectra of LA AR-3 mutant DETC at pH 6 at time delays listed in the legend and for excitation at 570 nm. The difference in the fluorescence profile between steady-state emission and FLUPS measurements is caused by the phase matching conditions adjusted for up-conversion of the blue part of the emission spectrum. (B) Comparison of the normalized fluorescence decays between WT, DETC and ARCH-5 (dots). Kinetics were averaged over 15 nm around the fluorescence maximum. Solid lines present the best result obtained for fit with a bi-exponential decay function.

response function) convoluted with two decaying exponents was used as a fitting function. A three-exponential fit was used for ARCH-5 kinetics to take into account the 0.4 ps rise of the fluorescence signal. Extracted decay time constants and their relative amplitudes are condensed in Table 1 along with calculated average fluorescence lifetimes for each of the

proteins. The amplitude-weighted average fluorescence lifetimes for DETC and ARCH-5 are 29 ps and 66 ps, respectively. This approach is explained and justified in the ESI (Section 2.2†). Note that they are proportional to the above-determined Φ_f of these proteins.

Additionally, to investigate the photoisomerization reaction and dynamics in the excited state of the proteins, ultrafast transient absorption spectroscopy (TAS) experiments were performed for the three proteins in their LA forms. In this method, the absorption changes in the 470–1000 nm range were monitored with 60 fs resolution after photoexcitation. The results of the TAS experiments are shown in Fig. 6. Panel A presents the time evolution of the difference absorption spectra of WT AR-3 (results for DETC and ARCH-5 are presented in the ESI†). We observe difference spectra typical for microbial retinal proteins, such as bR,^{29–32} highlighting excited state decay and isomerization on a sub-picosecond time scale. Indeed, at 50 fs the positive peak around 470–530 nm is due to excited state absorption (ESA) and the negative bands around 550 nm and 800–1000 nm range are assigned to the ground state bleaching (GSB) and stimulated emission (SE), respectively. Noise in the GSB region at 570 nm is caused by residual pump laser scattering. In the first 100 fs range ultrafast relaxation out of the Frank–Condon region takes place which explains the blue shift of the ESA and rise of SE in the near-IR. Later, along with the decay of the excited state (decay of ESA and SE) a new positive band appears around 630 nm. It reflects the all-*trans* to 13-*cis* isomerization of retinal, forming the vibrationally hot J intermediate, which in the next ps cools down (blue shift of its absorption to 610 nm) leading to the first ground state photoproduct of AR-3 (K intermediate).

The bottom part of Fig. 6 shows the comparison of the normalized ESA (panel B) and SE (panel C) dynamics of WT (red), DETC (blue) and ARCH-5 (yellow). Similar to the FLUPS results, DETC and ARCH-5 have significantly longer-lived excited states. At the short times (before 0.5 ps) the excited state kinetic traces of the mutants show an unusual sudden drop at 0.6 ps, which is due to excited state relaxation from the Franck–Condon region, but without the loss of its excited state population. Indeed, the stimulated emission remains constant during this time interval (see the complete time-dependent spectra for DETC and ARCH-5 in ESI,† panels A of Fig. S3 and S5†). In contrast with the WT, only a small absorption trace of photoproduct formation is detected for DETC and even weaker so for ARCH-5 for delays of hundreds of picoseconds, *i.e.* after excited state decay (Fig. S6†). The TAS results were globally fitted with three time constants. Results of the fit are presented in Table 2 and in panel C of Fig. S3–S5,† which represent the

Table 1 Fluorescence lifetimes, its relative amplitudes, amplitude-weighted average fluorescence lifetimes, measured fluorescence quantum yields and calculated non-radiative (k_{nr}) and radiative (k_r) rates of WT AR-3 and mutants at pH 6

Sample	τ_1 , ps	A_1 , %	τ_2 , ps	A_2 , %	$\langle\tau_{\text{fluo}}\rangle$, ps	Φ_f	k_{nr} , ns ⁻¹	k_r , ns ⁻¹
AR 3 pH 6	0.34 ± 0.02	85 ± 2	1.4 ± 0.1	15 ± 2	0.5		2000	
DETC pH 6	7.7 ± 1.1	37 ± 6	42 ± 3	63 ± 6	29	3.6 ± 0.6 × 10 ⁻³	34	0.12
ARCH-5 pH 6	31 ± 8	43 ± 9	91 ± 15	56 ± 3	66	9 ± 1 × 10 ⁻³	15	0.14



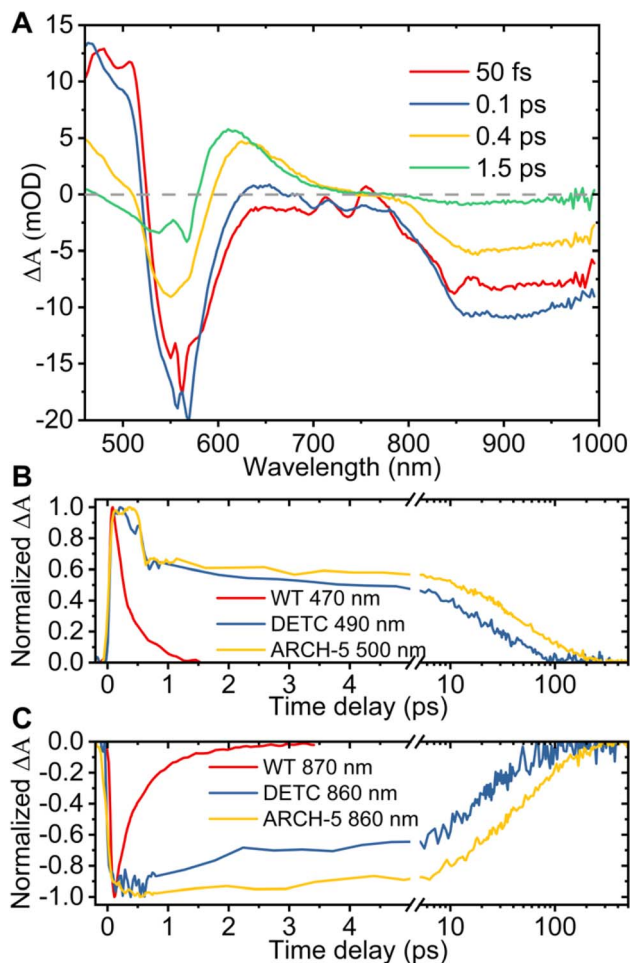


Fig. 6 Results of TAS experiments. (A) Difference absorption spectra of WT AR-3 at the time delays indicated in the legend after the 570 nm photoexcitation. Comparison of the normalized excited state absorption (B) and stimulated emission (C) kinetics at the indicated wavelengths for WT, DETC and ARCH-5 at pH 6.

decay-associated difference spectra for each of the samples, respectively. For WT AR-3, short-lived components τ_1 and τ_2 are mainly attributed to the excited state decay and formation of the hot photoproduct. The decay time of the photoproduct (K intermediate) is beyond the detection time range of these experiments. For the mutants, on the other hand, τ_1 is attributed to the thermal relaxation of the excited state and τ_2 and τ_3 to the excited state decay. Excited state lifetimes are consistent with the ones obtained in the FLUPS experiments. A direct comparison of the kinetic traces relevant to the excited state decay displays perfect agreement (not shown). The differences

Table 2 Global fitting results of TAS experiments

	WT AR-3	DETC	ARCH-5
τ_1 , ps	0.25 ± 0.02	0.4 ± 0.07	0.43 ± 0.03
τ_2 , ps	1.4 ± 0.2	7 ± 1	28 ± 5
τ_3 , ps	∞	42 ± 4	79 ± 6

in the value of τ_1 for WT AR-3 (Tables 1 and 2) are a consequence of the assumption of wavelength-independent lifetimes (global fit) for the TAS data.

2.4 Isomerization quantum yield

To determine the isomerization quantum yield of the WT AR-3 ($\phi_{\text{iso}}^{\text{AR-3}}$), we performed a nano-second transient absorption measurement and compared the bleach signal intensity with bR. $\phi_{\text{iso}}^{\text{AR-3}}$ is expressed as,³¹

$$\phi_{\text{iso}}^{\text{AR-3}} = \frac{\varepsilon_{\text{bR}}(\lambda') \Delta A_{\text{AR-3}}(\lambda)}{\varepsilon_{\text{AR-3}}(\lambda) \Delta A_{\text{bR}}(\lambda')} \phi_{\text{iso}}^{\text{bR}} \quad (1)$$

where $\varepsilon_{\text{bR}}(\lambda')$ and $\Delta A_{\text{bR}}(\lambda')$ are the molecular extinction coefficient and the bleach signal of bR at probe wavelength λ' , $\varepsilon_{\text{AR-3}}(\lambda)$ and $\Delta A_{\text{AR-3}}(\lambda)$ are the molecular extinction coefficient and the bleach signal of AR-3 at probe wavelength λ , respectively, and $\phi_{\text{iso}}^{\text{bR}}$ is the isomerization quantum yield of bR. The transient absorption spectra of light-adapted WT AR-3 were measured with 532 nm excitation wavelength (Fig. 7, panels A–C). At $t \sim 100 \mu\text{s}$, a blue-shifted M intermediate was accumulated. In this time region, the bleach signal of the initial state was observed without overlapping any absorption of red-shifted intermediates, e.g., K and O intermediates. To extract the bleaching signal, the transient absorption change was analyzed by a multi-exponential function assuming an un-branched sequential reaction. To observe the short-lived photo-intermediates, a measurement at specific probe wavelengths with better time resolution using a photomultiplier tube was performed (Fig. 7, panel C). Based on the result of the multi-exponential analysis, we constructed a photocycle model of the WT AR-3 (Fig. 7, panel D) and calculated the absorption spectra of two photointermediates appearing after $t = 35 \mu\text{s}$. Comparing the order of appearance and peak wavelengths, they were assigned to the quasi-equilibrium between the L and the M and between the N and the O intermediates, (Fig. 7, panel E).³² Because there is a weak contribution of the L even at the time point where the M accumulation is maximized, we estimated the bleach signal at 600 nm probe wavelengths where no absorption of the photointermediate overlaps with the bleach signal at $t \sim 100 \mu\text{s}$. To compare the bleach signal intensity between AR-3 and bR, the samples were excited with low pulse energy (0.50 mJ cm^{-2}) in the range where the linearity of the signal against the excitation pulse energy is maintained (Fig. S7, panel A†), and the absorptions of the WT AR-3 and bR sample solutions at the excitation wavelength were adjusted to the same with an error less than 1%. The transient absorption signals of the WT AR-3 and bR are shown in panel J of Fig. 7. $\Delta A_{\text{AR-3}}(600 \text{ nm})/\Delta A_{\text{bR}}(575 \text{ nm})$ in eqn (1) was determined to be 0.36. To estimate the $\varepsilon_{\text{AR-3}}(\lambda)$, we observed the absorption change along with the hydrolysis reaction of the Schiff-base linkage of the retinal chromophore (Fig. S7, panel B†). The absorption of retinal oxime produced by the hydrolysis reaction was observed at 362 nm. Based on the difference molecular extinction coefficient of the retinal oxime ($33 \text{ 600 M}^{-1} \text{ cm}^{-1}$ (ref. 33)) and the ratio between the absorption of the retinal oxime and the bleaching of the initial AR-3, the $\varepsilon_{\text{AR-3}}(600 \text{ nm})$ was determined



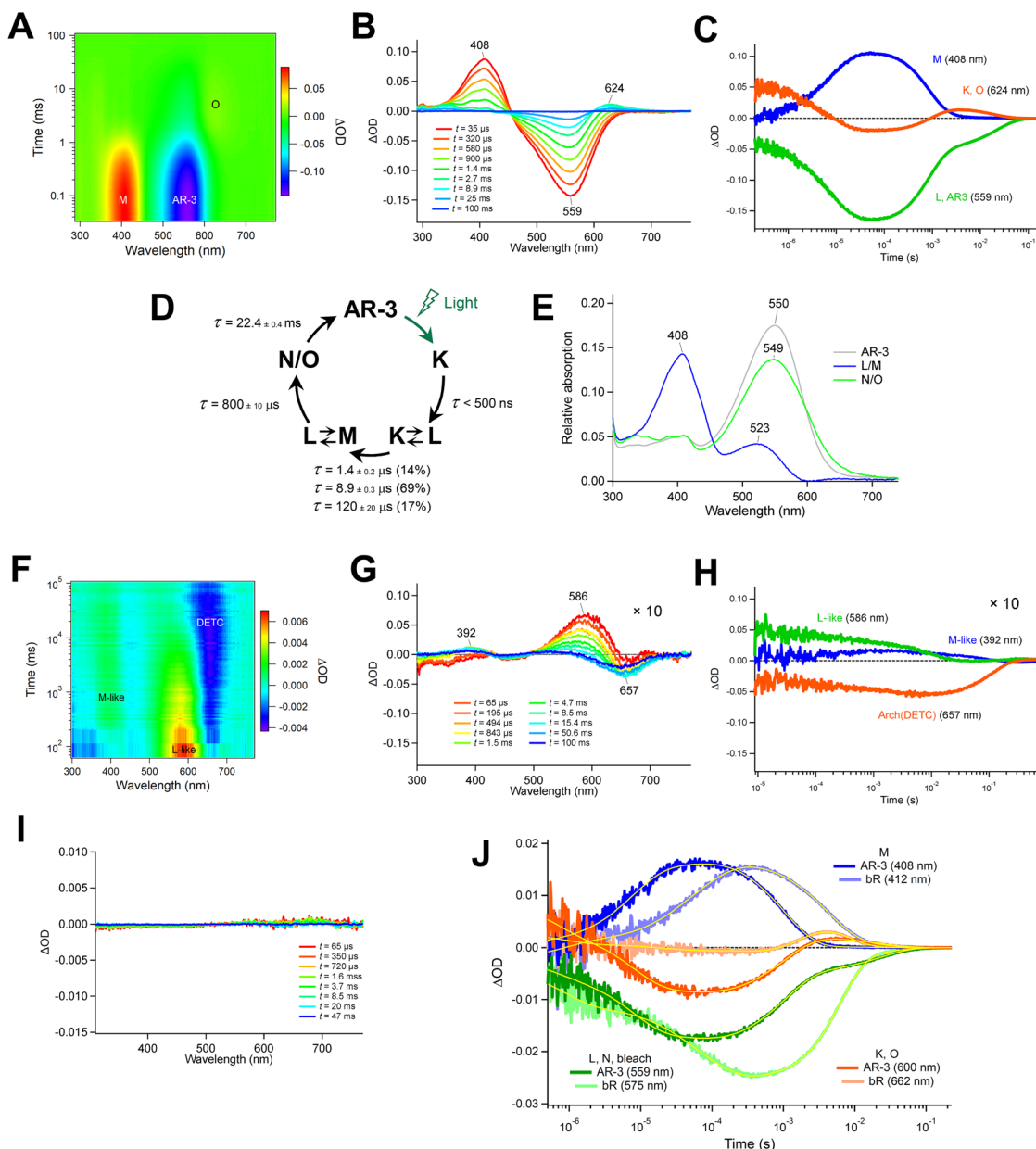


Fig. 7 Transient absorption measurements of the LA WT AR-3, DETC, and ARCH-5 in the microsecond to millisecond time region. (A and F) 2D plot of the transient absorption change of the WT AR-3 (A), DETC (F). (B, G, and I) Transient absorption spectra of WT AR-3 (B), DETC (G), and ARCH-5 (I) at specific time delay (t) after the photoexcitation. (C) Time-evolutions of the transient absorption change of the WT AR-3 probed at 408, 559, and 624 nm representing the M, the L/AR-3, and the K/O, respectively. (D) The photocycle model of the WT AR-3 obtained by the multi-exponential analysis of the transient absorption change. (E) The calculated absorption spectra of the photo-intermediates. (H) Time-evolutions of the transient absorption change of DETC probed at 392, 586, and 657 nm representing the M-like, the L-like, and the DETC, respectively. (J) Transient absorption signals of the WT AR-3 and bR with 0.50 mJ cm^{-2} excitation pulse energy. For clarity, the signal intensities of G and H have been amplified tenfold.

to be $24\,900 \pm 300 \text{ M}^{-1} \text{ cm}^{-1}$ ($\epsilon_{\text{AR-3}}(559 \text{ nm})$ was $48\,900 \pm 300 \text{ M}^{-1} \text{ cm}^{-1}$ at the peak maximum). Based on these values, $\phi_{\text{iso}}^{\text{bR}} = 0.64 \pm 0.04$,³⁴ and $\epsilon_{\text{bR}}(575 \text{ nm})$ $62\,000 \pm 1000 \text{ M}^{-1} \text{ cm}^{-1}$,³⁵ $\phi_{\text{iso}}^{\text{AR-3}}$ was determined to be 0.54 ± 0.03 .

Then, we investigated the photoreaction of DETC and ARCH-5 mutants. DETC excited at 600 nm exhibited two blue-shifted photointermediates at 586 and 392 nm (Fig. 7, panels F and G). Because the former appeared faster than the latter, whose absorption wavelength is in the near-UV region indicating to

have a deprotonated retinal Schiff-base, the 586 and 392 nm species were named L-like and M-like intermediates. It was difficult to determine the precise lifetimes of these intermediates due to the extremely small transient absorption signals of DETC. The signal intensity of the transient absorption change of DETC is less than 1/20 of the WT AR-3, indicating its isomerization efficiency is considerably smaller than that of the WT AR-3, *i.e.* in the range of 0.02–0.03. ARCH-5 excited at 600 nm did not exhibit any significant transient absorption change



(Fig. 7, panel I), indicating the photoisomerization reaction hardly occurs in this mutant.

We also searched for a photo-product signal on the sub-nanosecond time scale by performing femtosecond TAS. The K-like intermediate shows up in WT AR-3 as a pronounced absorption band peaking at 605 nm (Fig. 6A), red-shifted, as usual, with respect to the ground state absorption spectrum. Similar signatures hardly emerge from the noise for DETC and ARCH-5 in the range of 650–700 nm, and the very weak GSB is obscured by pump light scattering (570–600 nm, Fig. S6†). A quantitative evaluation of the quantum yield Φ_{iso} is therefore impossible, also due to the strong spectral overlap of GSB and the K-like absorption band. We note however, that these data are in agreement with the nano-second TAS (see above), in that ARCH-5 has a lower Φ_{iso} than DETC, and $\Phi_{\text{iso}}^{\text{AR-3}} > 10 \times \Phi_{\text{iso}}^{\text{DETC}}$.

The low Φ_{iso} values in the mutants support the conclusion that, relative to WT AR-3, the brighter steady-state emission of DETC and ARCH-5 at pH 6 is induced by a 1-photon absorption process, like in other AR-3-based GEVIs^{22,23} and in NeOR.⁹ Not only is the fluorescence intensity proportional to the excitation power, these mutants have a too low Φ_{iso} to form a putative fluorescent intermediate of sufficient amount, in stark contrast to WT AR-3. In addition, the fact that the ratio of the average fluorescence lifetimes agrees with the ratio of the Φ_{f} 's for both mutants is in strong support of the 1-photon excitation process. Interestingly, a comparison with the S_1 lifetime and Φ_{f} of NeOR⁹ shows that DETC and ARCH-5 have a radiative rate very similar to the former (*cf.* Table 1, $k_{\text{r}} \approx 1.7 \times 10^8 \text{ s}^{-1}$ for NeOR). Note however that in both mutants, the RPSB is present in a mixture of all-*trans*/15-*anti* and 13-*cis*/15-*syn* isomers, as demonstrated by both the HPLC analysis and Raman spectroscopy. Light-adaptation only enhances the proportion of the all-*trans*/15-*anti*, but it remains limited to 30%. Both fluorescence up-conversion and TAS experiments show a bi-exponential excited state decay, with relative amplitudes similar to the isomer composition (*cf.* Table 1). The most direct interpretation is then that the decay times τ_1 and τ_2 represent the decay of all-*trans*/15-*anti* and 13-*cis*/15-*syn* isomers, respectively (Fig. S9†). This assignment may be questioned since for other microbial rhodopsins, it was shown that the excited state lifetime of 13-*cis*/15-*syn* is shorter than the one of all-*trans*/15-*anti*.^{36–38} The assignment is therefore preliminary and will be substantiated through ongoing work comparing the temperature-dependant excited state decays of both mutants in light- and dark-adapted states. Unfortunately, since the absorption spectra of both all-*trans*/15-*anti* and 13-*cis*/15-*syn* isomers largely overlap (Fig. S8B and C†), it is impossible to determine the Φ_{iso} and Φ_{f} independently for one or the other isomer. The quoted values are thus an average over the relative amount both forms.

3 Computational results and discussion

3.1 Computed excited state dynamics of AR-3

We use an automatically constructed quantum mechanics molecular mechanics (QM/MM) model of AR-3 to simulate its

room-temperature photoisomerization dynamics at the population level (see details in Fig. 8A, B and the Method section†). The result provides a validation of the constructed QM/MM model as well as a basis for the assignment of part of the spectroscopy data presented above. Most importantly, it discloses different mechanistic aspects of the photoisomerization dynamics in this class of archaea-rhodopsins at both the geometrical and electronic structure levels.

As illustrated in Fig. 9A the simulated isomerization, occurs on a sub-picosecond timescale indicating the existence of a substantially barrierless path connecting the Franck–Condon (FC) (*i.e.* vertical excitation) region of the population to an intersection seam (IS_{S_1/S_0}) formed by CoIn points with values of α (*i.e.* of the C12–C13–C14–C15 dihedral, see Fig. 8C) spanning a *ca.* 70–120° range. By fitting the change in the S_1 population fraction as a function of simulation time (see Fig. 9B), it is possible to compute an τ_{S_1} of ~ 190 fs at the complete active space self-consistent field (CASSCF) level³⁹ (see ESI for details†). This should be compared with the average fluorescence lifetime result reported in Table 1 above, where a larger time constant of ~ 340 fs is determined. In Fig. 9C we also show how the isomerization coordinate is dominated by the C13=C14 torsional deformation α .

While a ~ 150 fs difference is significant when considering the short reaction time scale, we notice that this could, in part, depend on factors such as the limited accuracy of the PES and forces calculated at the CASSCF level that miss a significant portion of dynamic electron correlation energy. We will further discuss this point below. In Fig. 9A we also see that only part of the population reaches the photoproduct configuration (*i.e.* the K intermediate). By counting the number of reactive trajectories and calculating their fraction with respect to the total number of trajectories (reactive + unreactive), it is possible to estimate an Φ_{iso} that is 0.63 and, therefore, relatively close to the experimentally determined value of 0.54 ± 0.03 . In conclusion, both the calculated τ_{S_1} and Φ_{iso} seem to support a quantitatively correct use of the constructed AR-3 model for mechanistic studies.

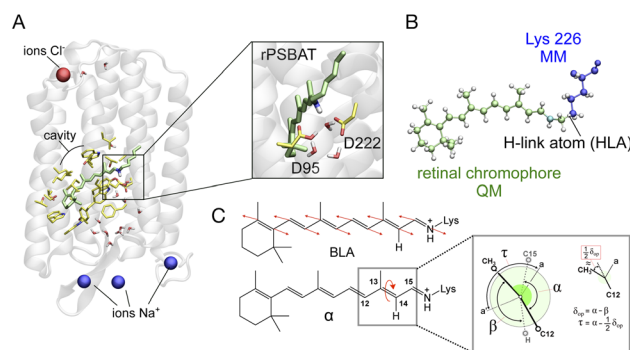


Fig. 8 (A) Structure of the QM/MM model of AR-3. An inset shows a magnification of the RPSB region. (B) The QM/MM partitioning and description of the treatment of QM/MM frontier is exemplified. (C) Schematic representation of the geometrical coordinates α , δ_{op} and BLA dominating the i_1 isomerization dynamics. The τ geometrical coordinate representing the orbital overlap across the π -bond is also defined.



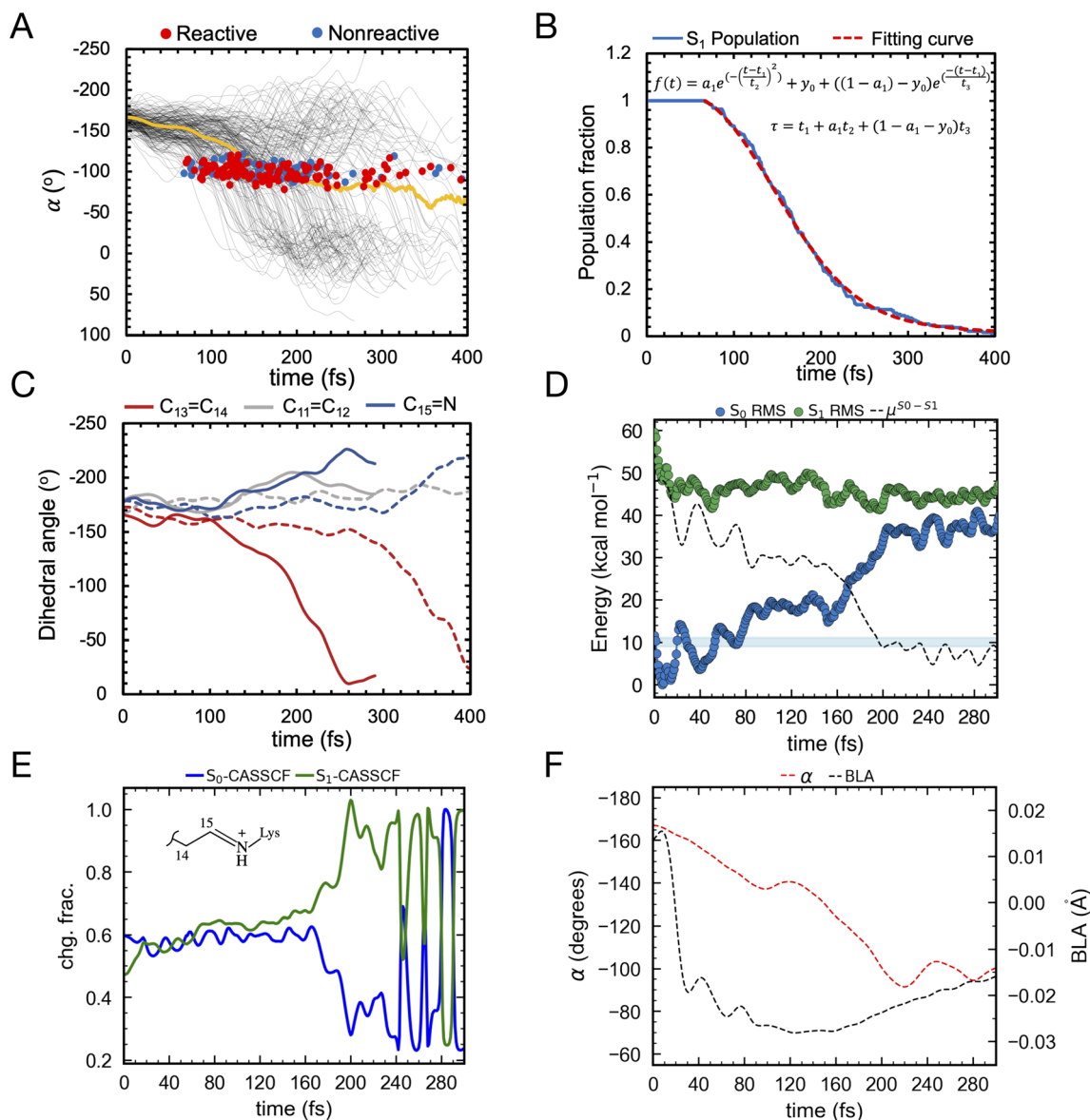


Fig. 9 (A) Evolution of α in all-*trans* AR-3 WT population dynamics. The evolution of α along S_1 trajectories (in black) and the corresponding average value (in yellow). Reactive (leading to isomerization) and non-reactive (aborted isomerization) decay points are represented by red and blue circles, respectively computed using 200 trajectories. (B) τ fitting from population dynamics (details in the ESI† and ref. 39). The S_1 population fraction as a function of time and the fitting function are given in blue and red, respectively. (C) Change in average dihedral angles of C11=C12 (ash), C13=C14 (red), and C15=N (blue) bonds across five randomly selected fast (solid lines) and five slow (dashed lines) trajectories. (D) Energy profiles along a single RMS-CASPT2 FC trajectory for AR-3 WT assuming zero kinetic energy. S_0 and S_1 profiles are reported as blue and green diamonds, respectively. The average S_1-S_0 energy is also given, while the horizontal blue bar indicates a 10 kcal mol⁻¹ threshold to emphasize the approaching of IS_{S_1/S_0} . (E) Evolution of the zeroth-order CASSCF charges associated to the AR-3 WT RMS-CASPT2 FC trajectory. The profiles correspond to the sum of such charges for the depicted moiety. (F) Evolution of α and BLA along the RMS-CASPT2 FC trajectory.

An effort has been made to calculate a more accurate τ_{S_1} value. To do so we used the state-of-the-art rotated multistate complete active space second-order perturbation theory (RMS-CASPT2)⁴⁰ for equilibrium structures and trajectory calculation. In fact, like CASSCF, RMS-CASPT2 correctly describes the intersecting S_1 and S_0 PESs near the CoIn but also accounts for the dynamic electron correlation missing in CASSCF energy and gradient computations.^{40,41} In spite of their availability, running hundreds of RMS-CASPT2 quantum-classical trajectories is still

highly unpractical. Therefore, we determined a scaling coefficient “ a_{scale} ”, relating CASSCF and RMS-CASPT2 gradients and resulting in a correction for the $\tau_{S_1}(t^{\text{RMS-CASPT2}} = a_{\text{scale}} \times t^{\text{CASSCF}}$, details in the ESI† and ref. 42). The resulting a_{scale} value of 1.25 produces an improved τ_{S_1} of 239 fs. This finding suggests that, as expected from previous studies⁴³ the inclusion of dynamic electron correlation leads to a flatter S_1 PES, or on the contrary, that the CASSCF method tends to produce systematically steeper excited state PESs.



To confirm the conclusion above, we were able to propagate a single RMS-CASPT2 quantum-classical trajectory released from the FC geometry with zero initial kinetic energy to approximately estimate the average dynamics of the population.⁴⁴ As shown in Fig. 9D, we found that such a trajectory indeed spans a shallower PES. The time necessary to approach the IS_{S_1/S_0} , that we conveniently define with a threshold of *e.g.* 10 kcal mol⁻¹ (see Fig. 9D) is significantly longer than in the corresponding CASSCF trajectory consistent with the scaling. However, a significant discrepancy between the computationally corrected τ_{S_1} and the experimental one remains, which is most probably due to our computations being based on a zero initial velocity single FC trajectory which does not properly account for the real ensemble of populations at room temperature.

3.2 Mechanistic aspects

In this section we look at some characteristics of the FC trajectory computed at the RMS-CASPT2 (and for comparison CASSCF) level of theory to discuss the general structural and electronic features of the S_1 relaxation of AR-3 towards the IS_{S_1/S_0} . We will then use the population dynamics simulated at the CASSCF level to discuss additional mechanistic features emerging from the population statistics. As shown in Fig. 9E, in the FC geometry the S_0 state of WT AR-3 is characterized by a mixed COV/CT electronic character, *i.e.* its wavefunction is described in terms of these “diabatic states” (see also next section). In more detail, the S_0 delocalized iminium positive charge is primarily (60%) residing in the chromophore portion incorporating the Schiff base moiety (*e.g.* with respect to the C13=C14 bond, see inset) and in S_1 away from the Schiff base moiety. This situation changes rapidly, after 160 fs, along the S_1 relaxation where the S_0 state becomes dominated by the CT character and the S_1 state by the COV character.

This rather distinct behavior with respect to, for instance, animal rhodopsins⁴⁵ has been previously reported for certain AR-3 variants⁴¹ but, also, for extremely red-shifted rhodopsins such as Neor.⁴⁶ As the C13=C14 bond twists, S_1 further increases in COV character until an α value of $\approx 90^\circ$ (see Fig. 9F) where the π -conjugation along the chromophore backbone is substantially broken and now S_1 features an almost pure COV character. The S_1 electronic character change is mirrored by less negative values of the BLA. In fact, as displayed in Fig. 9F, the BLA value (*i.e.* the difference between average single and average double bond lengths) increases as the COV character increases, along the trajectory.

As noted in a previous contribution by some of the authors,^{41,46} as the conjugation is broken, the COV character coincides with that of two unpaired electrons, thus it is more appropriate to reference this area of the S_1 PES as that of a short lived (or even transient) TIDIR as, indeed, a shallow S_1 energy minimum could be located. The COV character appears evident both by looking at the charges of the zeroth-order CASSCF and RMS-CASPT2 wavefunctions. In fact, this is naturally related to non-dynamic rather than dynamic electron correlation. Further inspection of Fig. 9D and F shows, consistently with existence of

a TIDIR species, a small trajectory segment where the S_1 energy, S_1 - S_0 energy gap and COV character do not change. In the case of the CASSCF wavefunction the nature of the S_1 electronic character oscillates revealing the presence of a nearby CoIn that, presumably, is not crossed in the RMS-CASPT2 description.

Going back to the CASSCF/6-31G* population dynamics, inspection of the geometrical changes associated with other relevant torsional degrees of freedom and accompanying the progression of α (see Fig. 8C) along the S_1 PES are documented in Fig. 9C. One can see that both the C15=N and C11=C12 bonds (measured using the skeletal dihedral angles C14-C15-N-C and C10-C11-C12-C13, respectively) undergo significant twisting in the counterclockwise direction relative to the clockwise α change and, thus, revealing a bicycle pedal motion active at the population level irrespective of whether the decay is slow or fast. On the other hand, such coupled motion is aborted upon decay as the “reactive” trajectories produce exclusively the 13-*cis* isomer of the chromophore upon S_0 relaxation. This is reminiscent of the isomerization coordinate of other animal⁴⁷ and microbial rhodopsins^{48,49} as originally proposed by Warshel.⁴⁷

Let us now discuss the vibrationally coherent character of the excited state reaction. Inspection of the distribution of the reactive “hop” events (see red line, Fig. 10A) displays, for delays of 100–250 fs, at least two reactive waves that are separated by a *ca.* 40–50 fs period. Importantly, a plot of the geometrical variable δ_{op} (see Fig. 10B) corresponding to an out-of-plane hydrogen wag at C14 somehow coupled with the wag of the heavier methyl substituent at C13, displays a similar periodicity indicating an impact of the δ_{op} wag phase at the time of the decay on the trajectory reactivity. Indeed, these data support the hypothesis that the so-called “HOOP mechanism” demonstrated in bovine rhodopsin is also active in this microbial rhodopsin whose isomerization occurs on the C13=C14 double bond. Thus the AR-3 simulation provides evidence that the replacement of a H-C11=C12-H moiety of animal rhodopsins with the Me-C13=C14-H moiety of microbial rhodopsins does not quench the HOOP mechanism for the control of trajectory reactivity (and, in turn, Φ_{iso} , see ref. 45) through a phase matching between the Me-C13=C14-H wag velocity and the α velocity. This implies that the larger is the population fraction decaying with a negative δ_{op} velocity at decay (the phase of α at decay is statistically positive due to the clockwise motion), the higher is the achieved Φ_{iso} value. Similar to the case of H-C11=C12-H wag in animal rhodopsins, the phase dependent reactivity is explained by a change in overlap velocity at the moment of decay.⁴⁵ This is illustrated by the compound geometrical variable τ (see Fig. 8C above), which is a combination of the variable describing the Me-C13=C14-H wag (HOOP) and the variable describing the C13=C14 isomerization (α), and that is proportional to the overlap between the fragment of π -orbitals whose interaction generates the C13=C14 π bond. The figure shows how the phase of the τ velocity is substantially dominated by the phase of the δ_{op} velocity (Fig. 10C) and that a positive τ velocity at decay marks reactive trajectories (see Fig. 10D).



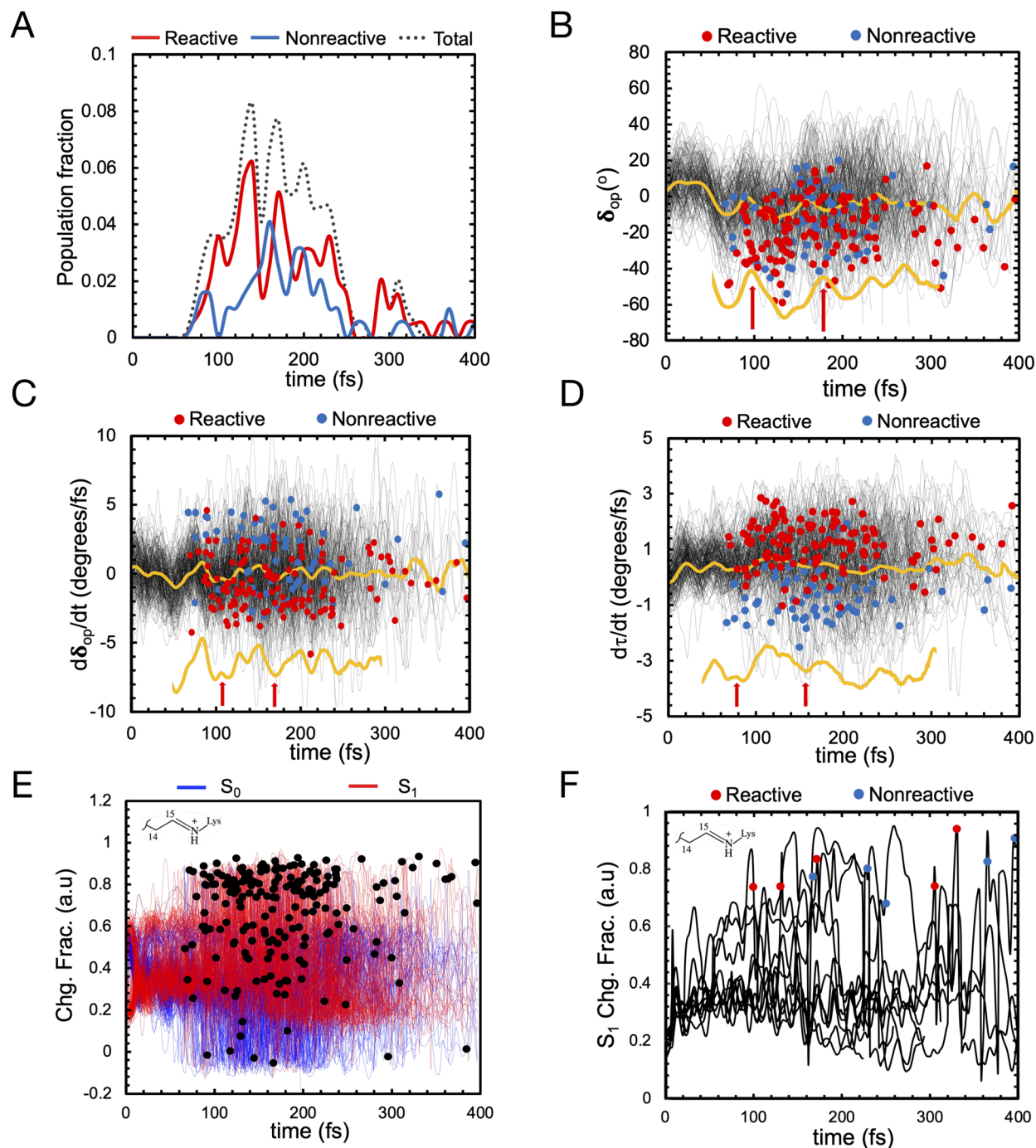


Fig. 10 (A) Oscillatory character of the S_1 total population decay (dotted line), reactive population decay (red) and non-reactive population decay (blue). (B) δ_{op} progression of S_1 population (in black) and corresponding average value (in yellow). Reactive (leading to isomerization) and non-reactive (aborted isomerization) decay points are represented by red and blue circles, respectively in figures B, C, D and F computed using 200 trajectories (C) δ_{op} progression velocity of S_1 population (black) and corresponding average value (yellow). (D) τ progression velocity of S_1 population (in black) and corresponding average value (in yellow). (E) Evolution of the zeroth-order CASSCF charges on the Schiff base moiety associated with S_0 (blue) and S_1 state (red). S_1 decay points represented in black dots. (F) Evolution of the zeroth-order CASSCF charge associated with S_1 state developed using ten randomly selected trajectories for the same moiety.

The above results convey a wealth of new information both on the S_1 dynamics as well as on the optogenetically relevance of WT AR-3 as a template for the development of novel fluorescent reporters (see next section). In fact, on one hand, the spectroscopy of LA WT AR-3 at pH 6, shows that both the

fluorescence decay kinetics and the ultrafast isomerization reaction monitored by TAS are consistent with several examples reported in the literature for other microbial rhodopsins, including BR.^{50–54} The dominant ≈ 300 fs decay time is due to the all-*trans*-to-13-*cis* isomerization process; an assignment



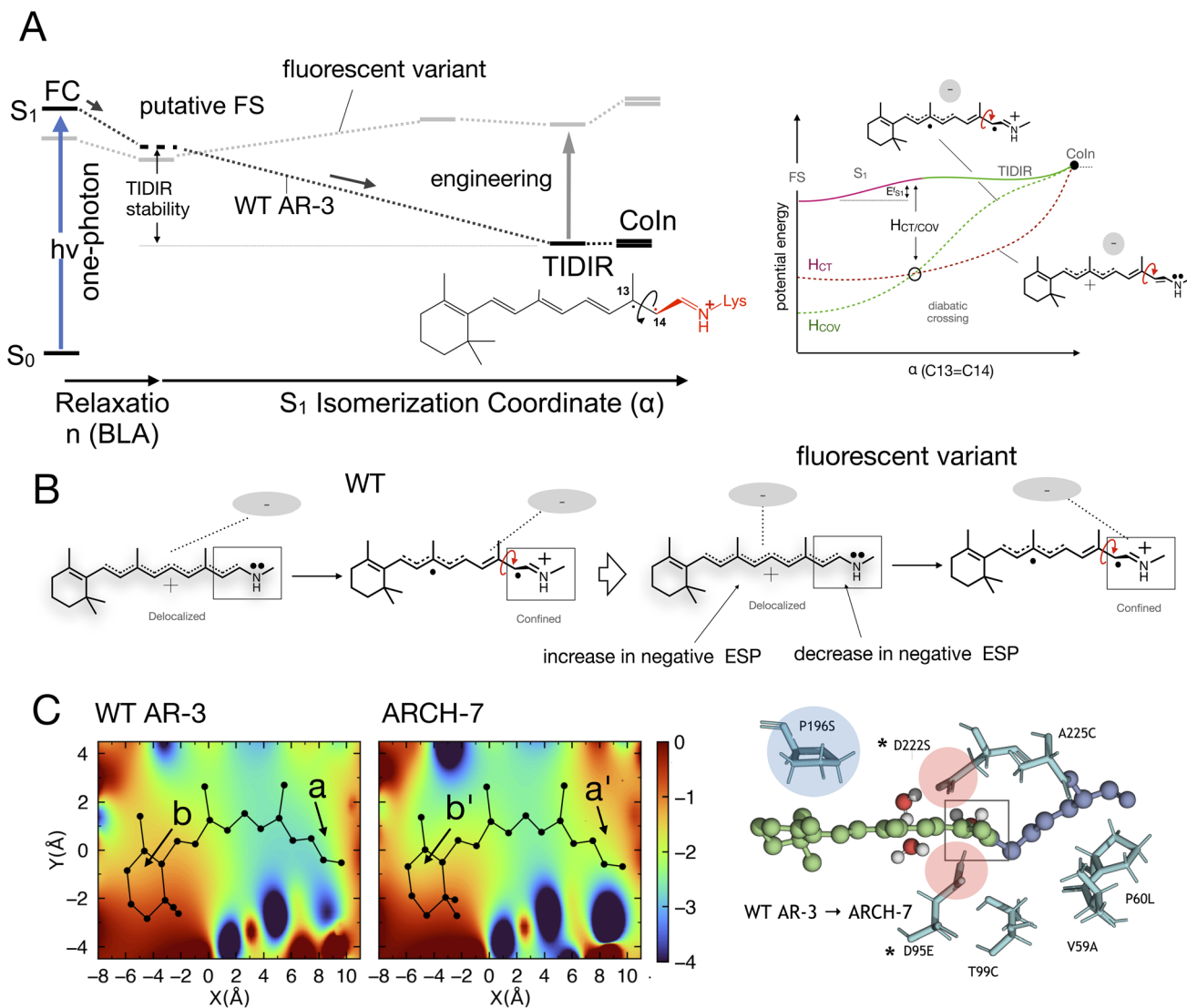


Fig. 11 (A) Schematic comparison between the WT AR-3 S_1 isomerization energy profiles (black) and the energy profile of a strongly emissive AR-3 variant (grey) (see also Fig. 1). It is evident that an engineering effort targeting high fluorescent AR-3 variants must destabilize the TIDIR region relative to the putative fluorescent state FS (bold vertical arrow). (B) Representation of the ESP changes through the displacement of a negative charge centroid when shifting from WT AR-3 to a fluorescent variant. (C) ESP cross-section computed for a WT AR-3 model and for the model of its ARCH-7 variant. A representation of the chromophore cavity mutation sites affecting the ESP is shown on the right (panel adapted from a previous work).¹¹ The position of two relevant carboxylic side chains is marked with a*.

supported by the QM/MM-based simulation of the light-induced population dynamics. In addition, these calculations also reveal that photoisomerisation occurs *via* a bicycle-pedal mechanism similar to the one found in animal rhodopsins (but occurring on the C13=C14-C15=N moiety rather than the C9=C10-C11=C12) and that Φ_{iso} is modulated by coherent oscillations of Me-C13=C14-H wagging motion, which is a feature of barrier-less and ballistic isomerisation as seen, again, in animal rhodopsins.⁴⁵

On the other hand, and in spite of the above similarities, it is shown here for the first time that WT AR-3 is electronically different from reference (*e.g.* bR) microbial rhodopsins and animal rhodopsins: it has an unusual degree of mixed COV and CT characters (described by the chromophore positive charge distribution. See Fig. 9E, 10E and F) in both the S_0 and S_1 states

already present in the Franck-Condon region and then conserved all along the reaction path. This and the related existence of a TIDIR intermediate close to the CoIn support the hypothesis that certain (presently unknown) fluorescent mutants of AR-3 may behave like Neor.^{11,46} However, WT AR-3 shows that the conclusion indicating that the an S_1 TIDIR with COV character necessarily implies an isomerisation-preventing S_1 barrier²⁰ is not valid any more. It is rather the destabilization of a pre-existing TIDIR “transient state” that modulates the barrier. It remains to be understood which exact electrostatic features of the RPSB-binding protein pocket WT AR-3, its mutants and in Neor lead to their unusual electronic properties, *i.e.* the formation of the S_1 TIDIR with COV character (*cf.* Fig. 11).



4 Engineering aspects

When considering the rational engineering of AR-3 towards more fluorescent variants, one has to deal with the overwhelming structural complexity of the cavity hosting the chromophore. Indeed, the fact that the cavity may incorporate the chromophore counterion in different positions and polar residues in different orientations in response to both electrostatic and steric factors, leads to patterns of electrostatic potential (ESP) and steric interactions whose effect is not obvious and very difficult to predict. For this reason, here we do not explicitly focus on the effect of amino acid replacements but, rather, on the changes in cavity electrostatics that, in principle, would lead to an increased excited state lifetime of DETC and ARCH-5 and other fluorescent AR-3 variants. This is assumed to be the most important factor for controlling the 1-photon Φ_r .

We start by noticing that the AR-3 population dynamics reported above shows that the S_1 electronic character changes accompanying the isomerization, are confirmed at the statistical level. Indeed, Fig. 10E and F show that the majority of trajectories display a COV character at the moment of S_1 decay (S_1 Schiff base charge fraction >0.8). In other words, the theory of photoisomerization in AR-3 mutants and Neor^{14,46} illustrated in Fig. 1 must also apply to WT AR-3. Again, this is different from the classical picture documented for bovine rhodopsin⁴⁵ and bacteriorhodopsin,⁵⁵ according to which S_1 has a CT character at decay (S_1 Schiff base charge fraction <0.2). It is this specific electronic character of the AR-3 chromophore that accounts for the impact of site-specific mutations on the variants' S_1 lifetime.^{14–16,44}

We now discuss which exact electrostatic features of the RPSB-binding protein pocket of WT AR-3 have to change to generate its brighter mutants such as DETC and ARCH-5 or, in perspective, variants as bright as Neor. As anticipated above, our reasoning relies on the formation of the S_1 TIDIR with COV character (*cf.* left panel in Fig. 11A). This implies that, (i) along the C13=C14 isomerization coordinate the diabatic states representing the CT and COV electronic characters cross (*cf.* right panel in Fig. 11A)⁴¹ and that the height of the crossing is proportional to the $E_{S_1}^f$ barrier, (ii) the cavity electrostatics acts according to a mechanism called “charge delocalization and confinement”⁴⁶ illustrated in Fig. 11B, and (iii) due to the initial mixed CT/COV character the FC and putative FS regions feature a chromophore with a “delocalized” positive charge while such charge is firmly “confined” in the C–C=N moiety in TIDIR (or the CoIn region accessed by the trajectories). The consequence is that an increase of negative ESP (or a decrease of positive ESP) in a chromophore cavity region located far from the C–C=N moiety must destabilize the TIDIR region and increase $E_{S_1}^f$ (see Fig. 11A and B). Due to lack of structural data, we could not computationally assess the above hypothesis for DETC and ARCH-5. Thus, in Fig. 11C we compare the ESP cross-section of WT AR-3 and its bright variant ARCH-7 whose QM/MM model has been reported.⁴¹ It can be seen that the electrostatic potential becomes more positive in the C–C=N moiety (compare a with a') region and more negative in the region close

to the chromophore ring (compare b and b') when going from WT to the variant.

Of course, these changes in ESP must be induced by the corresponding amino acid replacements that are shown on the right-end of the same figure. For instance, the replacements of the negatively charged oxygens of residue 222 with a single neutral serine oxygen combined with transferring the lost negative charge to residue 95 at a larger distance from the Schiff base, would contribute to projecting a less negative ESP on the C–C=N moiety. Similarly, the proline to serine replacement at position 196 could contribute to increasing the negative ESP far from the Schiff base. However, while one can attempt a rationalization of the ESP changes on the basis of the specific change in partial charge distribution and orientation induced by the replacements, the “inverse” problem of predicting which amino acid replacements would induce the wanted cavity ESP appears extremely complex and presently difficult to tackle if not in a highly hypothetical fashion. For this reason, we do not further elaborate on such point.

Regarding the voltage sensitivity of the fluorescence intensity, in AR-3, its D95N mutant, and Archon1, its origin is proposed to result from the voltage-dependent change in the hydrogen-bonding network. These changes are induced by shifts in the orientation of R92 or D125, which in turn affect the equilibrium between the protonated and deprotonated states of the RSB, leading to the change in the fluorescent state.^{1,22,56} Therefore, to improve the fluorescence quantum yields of AR-3 variants while maintaining their voltage sensitivity, amino acid residues not involved in the hydrogen bonding network around RSB, R82, and D125 should be targeted for mutation to increase the excited state barrier.

5 Conclusions

In the context of establishing WT AR-3 as an ideal template for engineering fluorescent GEVIs, we motivated the present combined experimental and computational study by two questions: does the TIDIR intermediate with COV character also exist in WT AR-3, and how to provide experimental evidence for the TIDIR-related PES barrier in mutants? The latter is addressed by a set of spectroscopic studies of the DETC and ARCH-5 mutants, which highlight the extended, barrier-induced excited state lifetimes of the all-*trans*/15-*anti* and 13-*cis*/15-*syn* isomers; a fact also found in other AR-3-based mutants.^{22,23} In addition, the larger excited state barriers in the mutants correlate with a largely reduced isomerization quantum yield, Φ_{iso} . The first question is addressed by an unprecedented non-adiabatic dynamics simulation of the WT AR-3 photoisomerization at the population level, which allows to compute both the excited state lifetime and Φ_{iso} thus providing a basis for a mechanistic interpretation of the experimental results. The most important message from the simulated population dynamics is that the WT AR-3 displays indeed a significant CT character in S_0 and a TIDIR intermediate with COV character in S_1 . This supports the hypothesis that the WT form of AR-3 has the same type of electronic character as the one found computationally in its weakly



fluorescent AR-3 variants as well as in the strongly fluorescent NeoR. However, WT AR-3 does not feature an S_1 isomerization barrier ($E_{S_1}^f$ is negligible or absent), and therefore decays rapidly to S_0 . Since other microbial rhodopsins failed to produce fluorescent variants with Φ_f 's similar to the ones obtained with AR-3 variants,¹⁵ we conjecture that the specific COV/CT mixture in S_0 and S_1 , as well as the presence of a TIDIR intermediate with full COV character, is the reason for the WT AR-3 propensity to form isomerisation-restraining S_1 barriers upon suitable genetic engineering.

The mechanism for the transition from WT AR-3 to a fluorescent form has been discussed and justified under the hypothesis that is controlled by variations in the cavity electrostatics and its interaction with the delocalized positive charge computed near the FC point and the TIDIR charge that is confined in the Schiff base moiety. This provides a rational route to engineering. In fact, it is possible to conclude that the problem of engineering highly fluorescent AR-3 variants, namely to increase the τ_{S_1} of the system, can be solved by finding mutations creating an ESP that destabilizes TIDIR with respect to FS. However, the problem of how to exactly identify the necessary mutation pattern (or group of different mutation patterns) remains to be further investigated.

Data availability

Additional data for this article, including HPLC, time-resolved spectroscopy, computational results, and the calculation of average fluorescence lifetimes are available in the ESI.†

Author contributions

The present paper is the result of a collaborative project of the teams led by M. O., M. S., K. I. and S. H. M. K. expressed and purified the proteins. K. H., M. K., I. W. and M. S. carried out the experiments and collected the data. D. W. and L. B. generated the QM/MM models and carried out all the calculations. All the authors contributed to analyzing, discussing, and interpreting the results, and to improving the different versions of the manuscript. K. H., D. W., M. K., L. B., M. O., K. I. and S. H. wrote the manuscript and ESI.†

Conflicts of interest

The authors declare no conflict of interest.

Acknowledgements

K. H and S. H. acknowledge the contributions of B. Marekha (ENS Lyon) in the early stage of the project, of O. Crégut for expert help in laser maintenance and development of data acquisition software and of L. Charbonnière (IPHC Strasbourg) for providing access to the near-IR Edinburgh fluorimeter. D. W., L. B. and M. O. are grateful to X. Yang (University of Siena) for his help in the computational part. Insightful discussions with the group of I. Moraes (NPL, London), and with N. Ferré and his team (Aix-Marseille University – CNRS) and with J.

Léonard are acknowledged as well. French partners are funded by the ANR project UltraArchae (grant no. ANR-21-CE11-0029). The Strasbourg team acknowledges financial support from ITI QMAT, funded by IdEx Unistra program (ANR 10 IDEX 0002). The Japanese partners are funded by JSPS KAKENHI Grants-in-Aid (grant number: JP23K05007 to M. K.; JP21H01875, JP23K18090, and JP23H04404 to K. I.), JST CREST (grant number: JPMJCR22N2 to K. I.), and MEXT Promotion of Development of a Joint Usage/Research System Project: Coalition of Universities for Research Excellence Program (CURE) (grant number: JPMXP1323015482 to K. I.). K. H, I. W. and M. S. thank the vibrational spectroscopy facility of the Advanced Characterization Platform of the Chevreul Institute, Lille (FR 2638). M. O. is grateful to the NSF CHE-SDM A for Grant No. 2102619. L. B. and M. O. are also grateful for partial support provided by EU funding within the MUR PNRR “National Center for Gene Therapy and Drugs based on RNA Technology” (Project no. CN00000041 CN3 RNA) – Spoke 6.

Notes and references

- 1 X. Meng, S. Ganapathy, L. van Roemburg, M. Post and D. Brinks, *ACS Phys. Chem. Au*, 2023, **3**, 320–333.
- 2 H. Kandori, *Bull. Chem. Soc. Jpn.*, 2020, **93**, 76–85.
- 3 J. Kralj, A. Douglass and D. Hochbaum, *Nat. Methods*, 2012, **9**, 90–95.
- 4 K. Kojima, A. Shibukawa and Y. Sudo, *Biochemistry*, 2020, **59**, 218–229.
- 5 R. S. McIsaac, M. K. M. Engqvist, T. Wannier, A. Z. Rosenthal, L. Herwiga, N. C. Flytzanis, E. S. Imashev, J. K. Lanyic, S. P. Balashov, V. Gradinarub and F. H. Arnolda, *Proc. Natl. Acad. Sci. U.S.A.*, 2014, **11**, 13034–13039.
- 6 N. C. Flytzanis, C. N. Bedbrook, H. Chiu, M. K. M. Engqvist, C. Xiao, K. Y. Chan, P. W. Sternberg, F. H. Arnold and V. Gradinaru, *Nat. Commun.*, 2014, **5**, 4894.
- 7 D. R. Hochbaum, Y. Zhao, S. L. Farhi, N. Klapoetke, V. K. Christopher A Werley, P. Zou, J. M. Kralj, N. S.-M. Dougal Maclaurin, J. L. Saulnier, G. L. Boulting, C. Straub, Y. K. Cho, M. Melkonian, G. K.-S. Wong, D. J. Harrison, V. N. Murthy, B. L. Sabatini, E. S. Boyden, R. E. Campbell and A. E. Cohen, *Nat. Methods*, 2014, **11**, 825–833.
- 8 K. D. Piatkevich, E. E. Jung, C. Straub, C. Linghu, D. Park, H.-J. Suk, D. R. Hochbaum, D. Goodwin, E. Pnevmatikakis, N. Pak, T. Kawashima, C.-T. Yang, J. L. Rhoades, O. Shemesh, S. Asano, Y.-G. Yoon, L. Freifeld, J. L. Saulnier, C. Riegler, F. Engert, T. Hughes, M. Drobizhev, B. Szabo, M. B. Ahrens, S. W. Flavell, B. L. Sabatini and E. S. Boyden, *Nat. Chem. Biol.*, 2018, **14**, 352–360.
- 9 M. Broser, A. Spreen, P. E. Konold, E. Peter, S. Adam, V. Borin, I. Schapiro, R. Seifert, J. T. M. Kennis, Y. A. Bernal Sierra and P. Hegemann, *Nat. Commun.*, 2020, **11**, 5682.
- 10 R. Palombo, L. Barneschi, L. Pedraza-González, D. Padula, I. Schapiro and M. Olivucci, *Nat. Commun.*, 2022, **13**, 6652.



- 11 L. Barneschi, E. Marsili, L. Pedraza-González, D. Padula, L. De Vico, D. Kaliakin, A. Blanco-González, N. Ferré, M. Huix-Rotllant, M. Filatov and M. Olivucci, *Nat. Commun.*, 2022, **13**, 6432.
- 12 D. Maclaurin, V. Venkatachalam, H. Lee and A. E. Cohen, *Proc. Natl. Acad. Sci. U.S.A.*, 2013, **110**, 5939–5944.
- 13 K. Kojima, R. Kurihara, M. Sakamoto, T. Takanashi, H. Kuramochi, X. M. Zhang, H. Bito, T. Tahara and Y. Sudo, *J. Phys. Chem. B*, 2020, **124**, 7361–7367.
- 14 S. Gozem, H. L. Luk, I. Schapiro and M. Olivucci, *Chem. Rev.*, 2017, **117**, 13502–13565.
- 15 M. d. C. Marín, D. Agathangelou, Y. Orozco-Gonzalez, A. Valentini, Y. Kato, R. Abe-Yoshizumi, H. Kandori, A. Choi, K.-H. Jung, S. Haacke and M. Olivucci, *J. Am. Chem. Soc.*, 2019, **141**, 262–271.
- 16 D. Agathangelou, P. P. Roy, M. del Carmen Marín, N. Ferré, M. Olivucci, T. Buckup, J. Léonard and S. Haacke, *C. R. Phys.*, 2021, **22**, 111–138.
- 17 J. Zhang, P. Singh, D. Engel, B. P. Fingerhut, M. Broser, P. Hegemann and T. Elsaesser, *Proc. Natl. Acad. Sci. U.S.A.*, 2024, **121**, e2319676121.
- 18 S. Schenkl, F. van Mourik, G. van der Zwan, S. Haacke and M. Chergui, *Science*, 2005, **309**, 917–920.
- 19 J. Léonard, E. Portuondo-Campa, A. Cannizzo, F. van Mourik, G. van der Zwan, J. Tittor, S. Haacke and M. Chergui, *Proc. Natl. Acad. Sci. U.S.A.*, 2009, **106**, 7718–7723.
- 20 M. Broser, T. Andruniów, A. Kraskov, R. Palombo, S. Katz, M. Kloz, J. Dostál, C. Bernardo, J. T. M. Kennis, P. Hegemann, M. Olivucci and P. Hildebrandt, *J. Phys. Chem. Lett.*, 2023, **14**, 9291–9295.
- 21 A. Penzkofer, A. Silapetere and P. Hegemann, *Int. J. Mol. Sci.*, 2019, **21**, 160.
- 22 A. Silapetere, S. Hwang, Y. Hontani, R. G. Fernandez Lahore, J. Balke, F. Velazquez Escobar, M. Tros, P. E. Konold, R. Matis, R. Croce, P. J. Walla, P. Hildebrandt, U. Alexiev, J. T. M. Kennis, H. Sun, T. Utesch and P. Hegemann, *Nat. Commun.*, 2022, **13**, 5501.
- 23 D. M. Nikolaev, V. N. Mironov, E. M. Metelkina, A. A. Shtyrov, A. S. Mereshchenko, N. A. Demidov, S. Y. Vyazmin, T. B. Tennikova, S. E. Moskalenko, S. A. Bondarev, G. A. Zhouravleva, A. V. Vasin, M. S. Panov and M. N. Ryazantsev, *ACS Phys. Chem. Au*, 2024, **4**(4), 347–362.
- 24 S. Oesterhelt, M. Meentzen and L. Schuhmann, *Eur. J. Biochem.*, 1973, **40**, 453–463.
- 25 S. O. Smith, M. S. Braiman, A. B. Myers, J. A. Pardoén, J. M. L. Courtin, C. Winkel, J. Lugtenburg and R. A. Mathies, *J. Am. Chem. Soc.*, 1987, **109**, 3108–3125.
- 26 S. O. Smith, J. A. Pardoén, J. Lugtenburg and R. A. Mathies, *J. Phys. Chem.*, 1987, **91**, 804–819.
- 27 J. Léonard, T. Gelot, K. Torgasin and S. Haacke, *9th International Conference on Photonics and Imaging in Biology and Medicine (Pibm 2010)*, 2011, vol. 277.
- 28 M. Gerecke, G. Bierhance, M. Gutmann, N. P. Ernsting and A. Rosspeintner, *Rev. Sci. Instrum.*, 2016, **87**, 053115.
- 29 K. Hasson, F. Gai and P. A. Anfinrud, *Proc. Natl. Acad. Sci. U.S.A.*, 1996, **93**, 15124–15129.
- 30 O. Bismuth, N. Friedman, M. Sheves and S. Ruhman, *J. Phys. Chem. B*, 2007, **111**, 2327–2334.
- 31 K. Inoue, Y. Sudo, M. Homma and H. Kandori, *J. Phys. Chem. B*, 2011, **115**, 4500–4508.
- 32 I. Chizhov, D. S. Chernavskii, M. Engelhard, K. H. Mueller, B. V. Zubov and B. Hess, *Biophys. J.*, 1996, **71**, 2329–2345.
- 33 B. Scharf, B. Pevec, B. Hess and M. Engelhard, *Eur. J. Biochem.*, 1992, **206**, 359–366.
- 34 J. Tittor and D. Oesterhelt, *FEBS Lett.*, 1990, **263**, 269–273.
- 35 S. Hashimoto, K. Obata, H. Takeuchi, R. Needleman and J. K. Lanyi, *Biochemistry*, 1997, **36**, 11583–11590.
- 36 A. Wand, R. Rozin, T. Eliash, K.-H. Jung, M. Sheves and S. Ruhman, *J. Am. Chem. Soc.*, 2011, **133**, 20922–20932.
- 37 A. Wand, N. Friedman, M. Sheves and S. Ruhman, *J. Phys. Chem. B*, 2012, **116**, 10444–10452.
- 38 A. Cheminal, J. Léonard, S.-Y. Kim, K.-H. Jung, H. Kandori and S. Haacke, *Phys. Chem. Chem. Phys.*, 2015, **17**, 25429–25439.
- 39 M. Manathunga, X. Yang, H. L. Luk, S. Gozem, L. M. Frutos, A. Valentini, N. Ferré and M. Olivucci, *J. Chem. Theory Comput.*, 2016, **12**, 839–850.
- 40 S. Battaglia and R. Lindh, *J. Chem. Theory Comput.*, 2020, **16**, 1555–1567.
- 41 Y. Nishimoto, S. Battaglia and R. Lindh, *J. Chem. Theory Comput.*, 2022, **18**, 4269–4281.
- 42 L. M. Frutos, T. Andruniów, F. Santoro, N. Ferré and M. Olivucci, *Proc. Natl. Acad. Sci. U.S.A.*, 2007, **104**, 7764–7769.
- 43 L. Barneschi, D. Kaliakin, M. Huix-Rotllant, N. Ferré, M. Filatov and M. Olivucci, *J. Chem. Theory Comput.*, 2023, **19**, 8189–8200.
- 44 S. Gozem, H. L. Luk, I. Schapiro and M. Olivucci, *Chem. Rev.*, 2017, **117**, 13502–13565.
- 45 X. Yang, M. Manathunga, S. Gozem, J. Léonard, T. Andruniów and M. Olivucci, *Nat. Chem.*, 2022, **14**, 441–449.
- 46 R. Palombo, L. Barneschi, L. Pedraza-González, D. Padula, I. Schapiro and M. Olivucci, *Nat. Commun.*, 2022, **13**, 6652.
- 47 A. Warshel, *Nature*, 1976, **260**, 679–683.
- 48 R. Palombo, L. Barneschi, L. Pedraza-González, X. Yang and M. Olivucci, *Phys. Chem. Chem. Phys.*, 2024, **26**, 10343–10356.
- 49 P. Altoè, A. Cembran, M. Olivucci and M. Garavelli, *Proc. Natl. Acad. Sci. U.S.A.*, 2010, **107**, 20172–20177.
- 50 R. A. Mathies, C. H. Brito Cruz, W. T. Pollard and C. V. Shank, *Science*, 1988, **240**, 777–779.
- 51 M. Du and G. R. Fleming, *Biophys. Chem.*, 1993, **48**, 101–111.
- 52 K. Hasson, F. Gai and P. Anfinrud, *Proc. Natl. Acad. Sci. U.S.A.*, 1996, **93**, 15124–15129.
- 53 J. Briand, J. Léonard and S. Haacke, *J. Opt.*, 2010, **12**, 084004–084016.
- 54 A. Wand, I. Gdor, J. Zhu, M. Sheves and S. Ruhman, *Annu. Rev. Phys. Chem.*, 2013, **64**, 437–458.
- 55 S. Gozem, P. J. Johnson, A. Halpin, H. L. Luk, T. Morizumi, V. I. Prokhorenko, O. P. Ernst, M. Olivucci and R. D. Miller, *J. Phys. Chem. Lett.*, 2020, **11**, 3889–3896.
- 56 E. C. Saint Clair, J. I. Ogren, S. Mamaev, D. Russano, J. M. Kralj and K. J. Rothschild, *J. Phys. Chem. B*, 2012, **116**, 14592–14601.

